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1 Accepted version before minor changes;

2 **Magma fragmentation in highly explosive basaltic eruptions induced by rapid crystallisation**

3

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25 **Basaltic eruptions are the most common form of volcanism on Earth and planetary bodies.**
26 **The low viscosity of basaltic magmas inhibits fragmentation, favouring effusive and lava-**
27 **fountaining activity, yet highly explosive, hazardous basaltic eruptions do occur. The**
28 **processes that promote fragmentation of basaltic magma remain unclear, and are subject to**
29 **debate. Here, we use a numerical conduit model to show that rapid ascent of magma during**
30 **explosive eruption produces large undercooling. Novel *in situ* experiments reveal that**
31 **undercooling drives exceptionally rapid (~minutes) crystallisation, inducing a step-change in**
32 **viscosity that triggers magma fragmentation. Experimentally-produced textures are**
33 **consistent with products of basaltic Plinian eruptions. We apply the numerical model to**
34 **investigate basaltic magma fragmentation over a wide parameter space and find that all**
35 **basaltic volcanoes have the potential to produce highly explosive eruptions. The critical**
36 **requirements are initial magma temperatures lower than 1100 °C, in order to reach a syn-**
37 **eruptive crystal content of > 30 vol.%, and thus a magma viscosity $\geq 10^5$ Pa s, which our**
38 **results suggest is the minimum viscosity required for the fragmentation of fast ascending**
39 **basaltic magmas. Our study provides both a demonstration and explanation of the processes**
40 **that drive basaltic Plinian eruptions, revealing how typically effusive basaltic volcanoes can**
41 **produce unexpected highly explosive, and hazardous, eruptions.**

42

43 Explosive eruptions are undoubtedly the most violent and catastrophic expression of volcanic
44 activity^{1,2}. Highly explosive Plinian eruptions can inject $> 0.1 \text{ km}^3$ of pyroclastic material producing
45 volcanic plumes that can reach up to 40 km in height, impacting both regional and global climate,
46 and producing a significant threat to proximal populations^{3,4}. Basaltic volcanoes are usually
47 characterised by effusive and mild-explosive lava fountaining activity⁵, however, basaltic Plinian
48 eruptions can occur⁶⁻¹¹. The main process characterising highly explosive activity is magma
49 fragmentation, which is the transition from ductile to brittle behaviour of molten magma^{1,2}. For
50 silicic magmas, fragmentation is attributed to high strain rates associated with acceleration of the
51 bubbly mixture^{1,12-14} or from the bubble overpressure associated with restricted bubble
52 expansion^{14,15-18}. The low viscosity of basaltic magmas, however, makes high strain rates, or the
53 bubble overpressure necessary to trigger fragmentation difficult to achieve^{1,19}.

54 Crystallisation during magma ascent may significantly increase magma viscosity, leading to
55 fragmentation²⁰. However, crystallisation in basaltic magmas has previously been thought to occur
56 on timescales significantly longer than the time required for magma to ascend from a crustal storage
57 chamber to the vent²¹, particularly in the case of highly explosive eruptions.

58 In volcanic conduits, the crystallisation kinetics of an ascending magma are driven by degassing
59 and cooling²²⁻²³. Plagioclase and pyroxene crystallisation are sensitive indicators of magma
60 dynamics in volcanic conduits^{21,24-27} and their kinetically controlled abundance can rapidly change
61 magma rheology²⁸⁻²⁹. Our understanding of crystallisation kinetics in magmas is underpinned by *ex*
62 *situ* crystallisation quench experiments. Here we perform novel *in situ* experiments in order to
63 visualize and quantify the evolution of rapid crystallisation in basaltic magmas, as with this
64 approach it is possible to ascertain in real time when crystals start to form and how quickly magma
65 crystallinity evolves.

66 A frequently used model to describe crystallisation as function of time is given by an exponential
67 law^{21,23}, where the rate of crystallisation is controlled by the characteristic time $\tau^{(c)}$. The
68 characteristic time is a measure of how fast a process will approach the equilibrium^{21,30}. The smaller

69 $\tau^{(c)}$, the faster crystals reach their equilibrium abundance (see Methods section). La Spina et al. (ref.
70 21) demonstrated that the time required to reach the equilibrium crystal fraction in basalts is ~ 4.6
71 times $\tau^{(c)}$. Furthermore, during mild lava-fountaining basaltic activity²¹, $\tau^{(c)}$ is in the order of
72 thousands seconds, resulting in a crystallisation time of ~ 2 hours. Crystallisation rate increases with
73 magma ascent rate, as cooling and decompression rates increase³⁰⁻³⁴. Therefore, $\tau^{(c)}$ is smaller and
74 the equilibrium crystal fraction will be reached faster. However, the characteristic times of crystal
75 growth during fast magma ascent have not been quantified.

76 In order to quantify the characteristic times of plagioclase and pyroxene crystallisation during
77 rapid ascent of basaltic magma, here we present the first *in situ* 4D (3D plus time) crystallisation
78 kinetics experiments under fast cooling rates, using fast synchrotron X-ray microtomography. Our
79 experiments provide the first estimation of the characteristic time for plagioclase and pyroxene
80 crystallisation in trachybasaltic magmas during a rapid and continuous increase of undercooling,
81 where ΔT is defined as the difference between the highest temperature at which plagioclase and
82 pyroxene is expected to crystallise and the temperature of the magma²⁴⁻²⁷.

83

84 **Crystallisation experiments at high undercooling**

85 Crystallisation experiments were performed *in situ* at beamline I12-JEEP, Diamond Light
86 Source, Harwell, UK, using a trachybasaltic glass (Supplementary Table 1) from the 2001 Etna
87 eruption as the starting material (see Methods). We combined a bespoke high-temperature
88 environmental cell³⁵ with fast synchrotron X-ray microtomography to image the evolution of
89 crystallisation in real time³⁶ in two experiments. In the first part of the experiment, crystallisation
90 was induced by decreasing temperature from 1250 °C to either 1170 °C or 1150 °C at 0.4 °C s⁻¹ at
91 ambient pressure with a dwell time at the final temperature of 4 h³⁶ (Supplementary Fig. 1). During
92 these 4 h, blocky and prismatic clinopyroxene and oxide crystals were able to grow³⁶ (Fig. 1 and
93 Supplementary Figs. 2 and 3), and the residual melt evolved to a basaltic trachyandesitic
94 composition (Supplementary Table 1; Supplementary Figs. 2 and 3). After 4 h at sub-liquidus

95 conditions (1170 °C and 1150 °C), the system was perturbed through a rapid cooling rate of 0.4 °C
96 s⁻¹, inducing a sudden and continuous increase of undercooling (ΔT). This produced a rapid
97 crystallisation event from the residual melt (Fig. 1; Supplementary Figs. 2 and 3). Our results show
98 that skeletal plagioclase crystals grew during this final stage of rapid cooling, specifically between
99 1112 and 1073 °C after a dwell time of 4 hours at 1150 °C (Fig. 1) and between 1131 and 1053 °C
100 after a dwell time of 4 hours at 1170 °C (Supplementary Fig. 3a,b,c). Plagioclase crystals grew to
101 equilibrium abundance in ~100 seconds (Fig.1), i.e. between two 3D scans. Following this initial
102 burst of rapid plagioclase growth, dendritic clinopyroxene crystals began to nucleate
103 heterogeneously on plagioclase and grew to their final size in the following 180 seconds (Fig. 1b,c;
104 Supplementary Figs. 2b,c, 3d,e, 4 and 5).

105 The large ΔT reached in a relatively short time during our *in situ* 4D crystallisation experiments
106 generated distinctive skeletal plagioclase crystals with swallow-tail morphology and dendritic
107 pyroxene (Fig. 2a-c), similar to the skeletal plagioclase and dendritic pyroxene crystals observed in
108 the products of explosive basaltic Plinian eruptions^{7-11,37,38}, such as Etna 122 B.C.^{7,8,37,38} (Fig. 2d).
109 Heterogeneous nucleation of dendritic pyroxene on skeletal plagioclase (Fig. 1b) is observed to
110 occur in ~180 s during our 4D experiments. These distinctive textures are also reported in products
111 of the Fontana Lapilli (Nicaragua)^{11,37,38} and 1886 Tarawera (New Zealand) eruptions^{7,9,10}.
112 Therefore, all the studied examples of basaltic Plinian deposits show features that are consistent
113 with the textures produced in our experiments. Furthermore, the signature skeletal and dendritic
114 pyroxene is also observed in sub-Plinian eruption (Yufune 2) products of Mt. Fuji (Japan)³⁹.

115 Skeletal plagioclase crystallisation occurred at ΔT between 60 and 140 °C with a growth rate of
116 3×10^{-5} cm s⁻¹ (mm h⁻¹) (Supplementary Table 2), whilst dendritic pyroxene crystallised at ΔT
117 between 60 and 175 °C with a growth rate of 2×10^{-5} cm s⁻¹ (mm h⁻¹) (Supplementary Table 2). This
118 indicates that a rapid increase of ΔT (>60 °C) induces fast crystallisation. As the equilibrium
119 pyroxene crystal content is achieved within ~180 s, we can infer that the pyroxene characteristic
120 time under large ΔT is < 40 s. For plagioclase, where the equilibrium crystal content is achieved

121 within 90 s the characteristic time is < 20 s. These are about two orders of magnitude less than the
122 characteristic time found by La Spina et al. (ref. 21) for effusive and mild lava fountaining activity
123 at Etna (Italy), Stromboli (Italy) and Kilauea (Hawaii), which involved much smaller ΔT (30-60
124 $^{\circ}\text{C}$)²¹ and magma ascent rates of ~ 3 m s⁻¹.

125 *Ex situ* experiments were also performed using a TZM cold seal pressure vessel apparatus, in
126 order to investigate whether fast crystallisation also occurs in a hydrous trachybasaltic melt during
127 rapid and continuous cooling and decompression (see Methods), simulating fast magma ascent in
128 the conduit. The experimental approach consisted of holding the sample for 30 minutes at 75 MPa
129 and 1070 $^{\circ}\text{C}$ before the sample was decompressed and the temperature decreased. After 30 minutes
130 pressure and temperature were decreased continuously for 300 s at 0.2 MPa s⁻¹ and 0.2 $^{\circ}\text{C}$ s⁻¹
131 respectively, reaching 15 MPa and 1010 $^{\circ}\text{C}$ (Supplementary Table 3; Supplementary Fig. 6). The
132 rapid decompression and cooling enabled large ΔT s to be achieved (>100 $^{\circ}\text{C}$; see
133 Supplementary Fig. 6) in 300 s, favouring predominantly clinopyroxene crystallisation, with minor
134 plagioclase and oxide (Supplementary Fig. 7). This indicates that $\tau^{(c)}$ is < 60 s, in agreement with
135 the order of magnitude estimated from *in situ* experiments.

136 Large undercooling can produce significant syn-eruptive microlite crystallisation during rapid
137 magma ascent²⁴⁻²⁷. This increase in crystallinity dramatically increases the viscosity of the
138 magma¹⁹. This process has been proposed to explain explosive basaltic Plinian eruptions, supported
139 by evidence of high microlite contents^{7-11,37,38}. However, a mechanism for this exceptionally fast
140 crystallisation during rapid magma ascent in basaltic Plinian eruptions, favouring fragmentation,
141 has not been demonstrated so far.

142

143 **Numerical simulation of the 122 B.C. Etna basaltic Plinian eruption**

144 Magma fragmentation in basaltic Plinian eruptions has been investigated with conduit models,
145 where crystallisation has been assumed either to be constant²⁰ or at equilibrium⁴⁰. Recent results
146 demonstrate that disequilibrium crystallisation plays a fundamental role in magma dynamics within

147 the conduit²¹, but syn-eruptive disequilibrium crystallisation has not yet been considered for basaltic
148 explosive volcanism^{20,40}.

149 We used the conduit model described by La Spina et al. (ref. 21, 23) to investigate the effect of
150 the new experimentally constrained characteristic times for crystallisation with large undercooling
151 derived here on the ductile-brittle transition of basaltic magma during highly explosive eruptions.
152 Since we are focussing our attention on explosive eruptions with high mass eruption rates ($>10^6$
153 kg/s), it is reasonable to assume that outgassing is negligible for this kind of activity. Indeed,
154 outgassing will be inhibited by fast magma ascent and fragmentation will be achieved faster than
155 outgassing⁴¹. As a test case, we consider the 122 B.C. Etna basaltic Plinian eruption⁶⁻⁸. To model
156 fragmentation we adopt the strain-rate criterion introduced by Papale (ref. 1):

$$157 \quad \dot{\gamma} = k \frac{G_{\infty}}{\mu} \quad (1)$$

158 where $\dot{\gamma}$ is the elongational strain rate, $k=0.01$ is a constant, μ is the magmatic viscosity and G_{∞}
159 is the elastic modulus at infinite frequency. The other constitutive equations are reported in the
160 Methods section. In Figure 3 we report the calculated plagioclase undercooling, crystal content and
161 viscosity as function of depth for $\tau^{(c)} = 10$ and 1000 s. Large undercooling is produced by adiabatic
162 expansion of exsolved volatiles, and mitigated by the latent heat of crystallisation for $\tau^{(c)} = 10$ s
163 (Fig. 3a). In this case, numerical results show a rapid increase in crystallinity and in viscosity at
164 depths below 2 km (Fig. 3c), leading to fragmentation and explosive Plinian eruption. Importantly,
165 numerical results indicate that strain-induced fragmentation is favoured by a combination of rapid
166 viscosity increase (about one order of magnitude in 5 seconds approaching the fragmentation depth)
167 and fast decompression/ascent rates (~ 0.2 MPa s^{-1} and ~ 15 m s^{-1}) that generate high strain rates. For
168 $\tau^{(c)} = 1000$ s, instead, the crystallisation rate is slow, and viscosity stays within the fragmentation
169 threshold all along the conduit.

170

171 **Sensitivity analyses on basaltic magma fragmentation driven by rapid crystallisation**

172 Having established that fast crystallisation plays an important role on basaltic Plinian eruptions,
173 we performed a sensitivity study with our numerical model to investigate the parameter space
174 whereby basaltic fragmentation driven by rapid crystallisation may occur. We use the Etna 122 B.C.
175 eruption as a test case. We focus on the behaviour of ΔT as a function of characteristic time of
176 crystallisation, pressure, temperature and magmatic H₂O content at the conduit inlet, conduit radius,
177 and initial phenocryst content. A detailed description of the initial condition of the sensitivity study
178 can be found in the Methods section. We performed several sensitivity studies assuming a
179 characteristic time of $\tau^{(c)} = 10$ s, as observed in our experiments, and $\tau^{(c)} = 1000$ s, as observed for
180 Etna 2001 in mild explosive activity. We also examined $\tau^{(c)} = 1$ and 100 s for completeness.

181 Numerical results show that undercooling is principally controlled by $\tau^{(c)}$ (Fig. 4a), because the
182 release of latent heat during rapid crystallisation (i.e. small $\tau^{(c)}$) affects the temperature of the
183 system, and, consequently, the undercooling. Therefore, undercooling and characteristic time are
184 linked, affecting one each other. However, our numerical results indicate that, within the parameters
185 space investigated, ΔT is always maintained between 60 and 190 °C (Fig. 4b), which is enough to
186 enable rapid crystallisation in any case. Furthermore, the sensitivity analyses indicate that, after the
187 characteristic time, initial temperature and water content (exsolved + dissolved) play an important
188 role on controlling undercooling (Fig. 4a). ΔT at the point of fragmentation increases as initial
189 temperature decreases (Fig. 4b), meaning that cooler magma in the chamber is more likely to
190 produce microlite crystallisation during ascent, as ΔT will already be large under the pre-eruptive
191 condition. An increase in the initial total H₂O (exsolved + dissolved) also produces an increase of
192 ΔT (Fig. 4c), caused by greater cooling of the system due to enhanced adiabatic gas expansion.
193 Furthermore, higher water content favours rapid crystallisation of microlites, due to the increase in
194 diffusivity within the melt.

195 Our sensitivity studies show that a lower initial temperature and a higher pre-eruptive crystal
196 content results in a greater likelihood of explosive eruptions (Fig. 4d,e). This is consistent with
197 estimates of pre-eruptive temperatures obtained for the Etna and Fontana Plinian eruptions, which

198 range between 1000 and 1060 °C³⁷. Regarding the pre-eruptive crystal content, however, products
199 erupted from basaltic Plinian eruptions are characterized by a small phenocryst content (<10
200 vol.%)^{7,8,10,11,37,38,40}. Therefore, in this context of low initial temperature and low phenocryst
201 content, our results highlight that the characteristic time of crystallisation has a fundamental role on
202 the likelihood of producing an explosive eruption. Indeed, a small $\tau^{(c)}$ is the primary cause for a
203 significant increase of the syn-eruptive crystal fraction over short timescales during magma ascent,
204 which consequently increases magma viscosity and, thus, the probability of magma fragmentation
205 (Fig. 4f). Moreover, our sensitivity analysis demonstrates that when the syn-eruptive crystal content
206 exceeds 30 vol.%, all numerical solutions reach the fragmentation threshold (Fig. 4f). Indeed, in a
207 fast ascending magma (from our sensitivity analyses we have an average ascent rate between ~5-50
208 m/s), this increase in syn-eruptive crystal content produces a rapid and dramatic increase of
209 viscosity (about one order of magnitude in less than 10 s close to the fragmentation depth),
210 increasing the likelihood of magma fragmentation. Numerical results also show that, as soon as
211 viscosity exceeds 10⁵ Pa s, all the numerical solutions reach the fragmentation threshold, generating
212 an explosive eruption (Fig 4g). Indeed, for a fast ascending magma where outgassing can be
213 neglected and strain-rates are high, the key parameter controlling the triggering of fragmentation is
214 viscosity. Therefore, 10⁵ Pa s represents a minimum viscosity determining a drastic change in
215 eruptive style for activities characterised by high mass eruption rates. This minimum viscosity is
216 one order of magnitude lower than previously reported for low-viscosity magmas^{1,42}.

217 Experimental and natural observations combined with a numerical model allow us to conclude
218 that pre-eruptive temperatures <1100 °C favour the formation of highly explosive basaltic
219 eruptions, such as Plinian volcanism, driven by fast syn-eruptive crystal growth under high
220 undercooling. This implies that all basaltic systems on Earth have the potential to produce powerful
221 explosive eruptions.

222 **METHODS**

223 **Starting material.** The starting material, used for our crystallisation experiments, is a
224 trachybasalt from the lower vents of the 2001 Mt. Etna eruption^{36,43}. The anhydrous, glassy starting
225 material was obtained by melting crushed rock samples in a Pt crucible. Melting was performed in a
226 Nabertherm® MoSi₂ box furnace at 1400 °C and at atmospheric pressure. The melt was left in the
227 furnace for four hours to allow the melt to fully degas and to dissolve the crystals present. The melt
228 was then quenched in air to glass. This procedure was repeated two times to homogenise the melt.
229 Finally, anhydrous glassy cylinders 3 mm in diameter and 4 mm in length were drilled from the
230 synthesized glass for *ex situ* and *in situ* 4D crystallisation experiments.

231 The chemical composition of the anhydrous glassy starting material has been analysed with a
232 Jeol JXA 8530 F microprobe at the Photon Science Institute, University of Manchester, Manchester,
233 UK, and are reported in Supplementary Table 1. Analyses were performed using a 15 kV
234 accelerating voltage, 10 nA beam current and beam size of 10 µm. Standards used for calibration
235 were albite for Na, periclase for Mg, corundum for Al, fayalite for Fe, tephroite for Mn, apatite for
236 P, sanidine for K, wollastonite for Ca and Si and rutile for Ti. Sodium and potassium were
237 measured first to minimize loss owing to volatilisation.

238 Hydrous trachybasaltic glass with ~3 wt. % of H₂O was obtained melting the starting material
239 and homogenizing H₂O in a Pt capsule at 100 MPa and 1200 °C. The hydrous starting glasses were
240 produced using a TZM cold seal pressure vessel apparatus at the School of Earth Sciences,
241 University of Bristol, Bristol, UK. Before to perform crystallisation experiments Raman spectra
242 were collected from the hydrous starting glass, using a Thermo Scientific™ DXRTMxi Raman
243 Imaging Microscope at the School of Earth Sciences, University of Bristol (Bristol, UK) in order to
244 check that the amount of H₂O was ~3 wt. %.

245

246 ***Ex situ* decompression and cooling experiments.**

247 *Ex situ* experiments were performed in order to investigate the process of fast crystallisation in a
248 hydrous trachybasaltic melt during a continuous decompression (0.2 MPa s^{-1}) and cooling ($0.2 \text{ }^\circ\text{Cs}^{-1}$)
249 ¹), simulating magma ascent in the conduit. Two experiments were performed using a TZM cold
250 seal pressure vessel apparatus at the School of Earth Sciences, University of Bristol, Bristol, UK.
251 The redox condition of the apparatus was NNO, adding ~1 vol.% of hydrogen to the pressuring
252 argon. Capsules ($\text{Au}_{80}\text{-Pd}_{20}$) were loaded with hydrous glass with a cylindrical shape. The
253 experimental approach consisted of holding the sample for 30 minutes at 75 MPa and 1070 °C
254 before being decompressed and cooled. The first experiment was quenched after 30 minutes at 75
255 MPa and 1070 °C, in order to texturally characterise the initial conditions before that fast
256 decompression and cooling were applied. In the second experiment, after 30 minutes pressure and
257 temperature were released continuously in 300 s at 0.2 MPa s^{-1} and $0.2 \text{ }^\circ\text{Cs}^{-1}$ respectively, reaching
258 15 MPa and 1010 °C (Supplementary Table 3; Supplementary Fig. 6). The sample was rapidly
259 quenched at the final conditions. Final pressure was chosen as approximate fragmentation pressures
260 calculated by modeling. Decompression and cooling rates were predicted from the conduit model
261 for magma ascent during basaltic Plinian eruptions.

262

263 ***In situ* synchrotron X-ray microtomography experiments.** The experiments were performed
264 at the beamline I12-JEEP⁴⁴, Diamond Light Source, Harwell, UK. For these *in situ* crystallisation
265 experiments, we used the high-temperature resistance Alice furnace⁴⁵, which allows us to control
266 cooling at $0.05 \text{ }^\circ\text{Cs}^{-1}$ to $0.4 \text{ }^\circ\text{Cs}^{-1}$, and the P2R *in situ* rig³⁵ for high speed rotation. Temperature was
267 measured with a R type thermocouple, which was close to the sample, positioned in the middle of
268 the hot spot. The dimension of the Alice furnace's hot spot with homogeneous temperature is
269 ~5x5x5 mm; our samples were positioned within this area. The R type thermocouple provided us
270 the temperature of the sample with an uncertainty of $\pm 5 \text{ }^\circ\text{C}$. The sample holder was an alumina
271 crucible (Supplementary Fig. 2), which is suitable for the temperature range investigated and it has
272 low X-ray attenuation coefficient. The glassy cylinders were heated *in situ* in the Alice furnace up

273 to 1250 °C for 30 minutes (Supplementary Figure 1). After the initial annealing period,
274 crystallisation was induced by decreasing temperature from 1250 °C to 1170 °C or 1150 °C at
275 ambient pressure, holding at the final temperature for 4 h³⁶ (Supplementary Figure 1). After this
276 step, the system was perturbed by a rapid cooling at rate of 0.4 °Cs⁻¹ in order to investigate the rapid
277 crystallisation in real time (Supplementary Figure 1), reaching high undercooling (up to ~180 °C) in
278 a short time.

279 The experiments were performed in phase-contrast mode, setting the sample-to-detector distance
280 at 2300 mm in order to work in the edge-detection regime⁴⁶ (Supplementary Table 4). The
281 projections were acquired using a monochromatic X-ray beam with energy of 53 keV. In each scan,
282 1800 tomographic projections were acquired by the detector with equiangular steps over a full
283 rotation angle of 180° (Supplementary Table 4). The exposure time for the acquisition of each
284 projection was 0.05 s (Supplementary Table 4), therefore, the temporal resolution of each scan was
285 of 90 seconds. The isotropic pixel size is 3.2 µm. The detector was a high-resolution imaging PCO
286 edge camera with optical module 3, corresponding to a field of view of 8.0 mm × 7.0 mm. Scan
287 acquisition started before the end of the annealing, covering the cooling period between 1250 °C
288 and the dwell temperatures (1170 and 1150 °C), the entire duration of the dwell time and the final
289 rapid cooling rate of 0.4 °Cs⁻¹.

290

291 **Image reconstruction and processing.** Tomographic projections were reconstructed into 32-bit
292 slices by using Diamond I12 in-house python codes, using the *gridrec* algorithm^{47,48}
293 ([http://confluence.diamond.ac.uk/display/I12Tech/Reconstruction+](http://confluence.diamond.ac.uk/display/I12Tech/Reconstruction+scripts+for+time+series+tomography)
294 [scripts+for+time+series+tomography](http://confluence.diamond.ac.uk/display/I12Tech/Reconstruction+scripts+for+time+series+tomography))^{49,50}. The pre-processing pipeline includes centre of rotation
295 calculation⁴⁹, zinger removal, blob removal⁵⁰, and regularisation-based ring removal⁵¹.

296 The reconstructed slices were converted to 8-bit raw format and stacked using ImageJ software⁵²
297 to obtain volumes in which the isotropic voxel has an edge size of 3.2 µm. Reconstructed volumes
298 of experiments ET1150 and ET1170 were then cropped using Avizo® software v.8.0 (FEI

299 Visualization Sciences Group) in order to select the volume of interest (VOI) (Supplementary Table
300 4). In the experiment ET1150 plagioclase and pyroxene crystals nucleated and grew in a relatively
301 large pocket of melt (Figs 1 and 2). Therefore, the VOI selected consists of a volume of melt where
302 the rapid crystallisation of plagioclase and pyroxene occurred during the final rapid cooling rate of
303 $0.4\text{ }^{\circ}\text{Cs}^{-1}$. In the experiment ET1170 plagioclase and pyroxene crystals formed in narrow layers of
304 melt (Supplementary Fig. 3), during rapid continuous cooling at a rate of $0.4\text{ }^{\circ}\text{Cs}^{-1}$.

305 Three-dimensional visualization (volume rendering) of the reconstructed volumes was obtained
306 using the commercial software VGStudio 3.0 (Volume Graphics), which allowed us to make 3D
307 textural observations of the plagioclase and pyroxene crystal morphologies (Fig. 2). Therefore, the
308 reconstructed volume of each scan allowed us to quantify when and at which range of temperature
309 plagioclase and pyroxene crystals were able to form.

310

311 **Image segmentation and analysis of plagioclase.** Segmentation is the process that allows
312 separation of objects from the background to obtain binary volumes containing only the feature of
313 interest. Segmentation of plagioclase crystals from the glassy matrix was performed using the semi-
314 automatic volume segmentation^{53,54} in Avizo software v. 8.0 (Supplementary Table 4). This
315 segmentation requires manual drawing of the outlines of crystals on the 2D slices. This is repeated
316 every 5–10 slices, depending on the size of the crystal and the complexity of their shape, along the
317 crystal length. The crystal shape is reconstructed automatically by the software through an
318 interpolation procedure. The advantage of this technique is that the operator can obtain the real
319 morphology of the object of interest by visual inspection^{53,54}.

320 The reconstructed 3D images were processed and analysed with the Pore3D software library,
321 custom-developed at Elettra⁵⁵. The Pore3D software allowed us to quantify the number of
322 plagioclase crystals, the volume and the maximum length of each crystal, operating directly in the
323 3D domain^{55,56}. As we were able to obtain the 3D shape of plagioclase crystals and the real
324 maximum axis length (L_{3D}) we could calculate the growth rate (Y_{L3D}) of plagioclase crystals

325 (Supplementary Table 2), using the experimental duration of growth (experimental duration). The
326 growth rate was estimated using the following equation⁵³:

327
$$Y_{L3D} = (L_{3D} * 0.5) / t_{\text{growth}}$$

328 where t_{growth} is the time required for crystal growth. The microtomography images give us the
329 opportunity to measure the volume of crystals. The volumetric growth rate (Y_V) was calculated
330 (Supplementary Table 2) using the following relationship⁵³:

331
$$Y_V = (V * 0.5) / t_{\text{exp}}$$

332 where V is the volume of the crystal.

333

334 **Image analysis of pyroxene growth kinetics.** Back-scattered electron (BSE) images were
335 collected, using a JEOL JSM-6390LA FE-SEM at the School of Earth and Environmental Sciences,
336 University of Manchester, Manchester, UK, in order to analyse the pyroxene morphologies and
337 kinetics. We used an acceleration voltage of 15 kV and beam current of 10 nA. The sizes of
338 dendritic pyroxene crystals were measured in the 2D domain, using BSE images and ImageJ
339 software⁵², as pyroxene morphologies formed during continuous cooling are difficult to resolve and
340 analyse in the 3D domain. The pyroxene growth rate is calculated by dividing the entire length of
341 the dendritic crystal over the duration of the pyroxene growth (Supplementary Table 2), as dendritic
342 crystals grow in one direction.

343

344 **Electron microprobe analysis.** Samples obtained with *ex situ* and *in situ* crystallisation
345 experiments were analysed with a JEOL JXA-8530F field emission electron microprobe at the
346 Photon Science Institute, University of Manchester, Manchester, UK. For both instruments, the
347 operating conditions were as follows: 15 kV accelerating voltage, 10 nA beam current, and beam
348 diameter of 10 or 5 μm (the latter for microlites). Sodium and potassium were measured first to
349 minimise loss by volatilisation. Calibration standards were albite for Na, periclase for Mg,
350 corundum for Al, fayalite for Fe, tephroite for Mn, apatite for P, sanidine for K, wollastonite for Ca

351 and Si and rutile for Ti. Several compositional maps, with a nominal resolution of 1 μm^2 , have been
352 also collected using the EPMA (EPMA map, are reported in Supplementary Figs. 4, 5 and 7).

353

354 **Constitutive equations for the conduit model.** In this work we use the 1D steady-state model
355 for magma ascent described by ref. (21, 23, 57). The governing equations used in this work are
356 reported in ref. (57). The application to a specific volcano is achieved by providing constitutive
357 equations to describe the specific rheological, solubility, crystallisation, outgassing, and
358 fragmentation behaviour of the system.

359 Following ref. (58), the viscosity of the liquid phase is modelled as:

360
$$\mu_l = \mu_{melt} \cdot \theta(x_c^l),$$

361 where μ_{melt} is the viscosity of the bubble-free, crystal-free liquid phase and θ is a factor which
362 increases viscosity attributed to the presence of crystals⁵⁹.

363 We use an empirical relationship to estimate μ_{melt} as a function of water concentration and
364 temperature, as in ref. (60) (based on the Vogel-Fulcher-Tammann equation):

365
$$\log(\mu_{melt}) = A + \frac{B(y, x_{d_{H_2O}}^{md})}{T - C(y, x_{d_{H_2O}}^{md})},$$

366 where the viscosity μ_{melt} is in Pa s and T is the temperature in Kelvin. The parameter A is the
367 logarithmic value of the viscosity at infinite temperature and it is assumed to be constant for all
368 melts. The parameters B and C, instead, are functions of the melt composition y and of the dissolved
369 water content $x_{d_{H_2O}}^{md}$. In this work, we used the composition of the average melt inclusion
370 composition (Etna 122 B.C.) from ref. (61). Furthermore, as crystallisation proceeds, viscosity is
371 increased according to the empirical model described in ref. (62):

372
$$\theta = \frac{1 + \varphi^\delta}{[1 - F(\varphi, \xi, \gamma)]^{B\varphi^{*\delta}}},$$

373 where

374
$$F = (1 - \xi) \operatorname{erf} \left[\frac{\sqrt{\pi}}{2(1 - \xi)} \varphi(1 + \varphi^\gamma) \right], \quad \varphi = \frac{\left(\sum_{j=1}^{n_c} x_{c_j}^l \right)}{\phi^*}.$$

375 The fitting parameters B, δ, ξ, γ and ϕ^* chosen for this work are the same used in ref. (63).

376 The model proposed in this work takes into account two different gas components: water and
 377 carbon dioxide. The equilibrium profile of the dissolved gas content $x_{d_i}^{md,eq}$ of component i follows
 378 the Henry's Law, i.e.

379
$$x_{d_i}^{md,eq} = \sigma_i \left(\frac{P_{g,i}}{\bar{P}} \right)^{\varepsilon_i},$$

380 where $P_{g,i} = \alpha_{g_i} P_g / \alpha_g$ is the partial pressure of the i -th gas component expressed in Pa, $\bar{P} = 1$
 381 Pa is used to make the expression in the brackets adimensional, σ_i is the solubility coefficient and ε_i
 382 is the solubility exponent. We assume that the solubility parameter σ_i and ε_i are constant during
 383 ascent. For this work we adopted the following parameters $\sigma_{H_2O} = 1.8911 \times 10^{-6}$; $\varepsilon_{H_2O} = 0.5257$;
 384 $\sigma_{CO_2} = 2.2154 \times 10^{-12}$; $\varepsilon_{CO_2} = 1.075$. In this work, we assume also equilibrium exsolution, which
 385 means that the dissolved volatile contents always follow the equilibrium profile.

386 The crystallisation model adopted here has been proposed in ref. (23). We consider the three
 387 different major crystal components erupted by Etna volcano: plagioclase, pyroxene and olivine. We
 388 assume that crystals stay coupled with the melt (i.e. no fractional crystallisation). For a better
 389 modelling of crystal nucleation and growth, we also assume that the equilibrium crystal contents are
 390 functions of temperature, pressure and dissolved water content. With these assumptions, the
 391 equilibrium mass fraction $x_{c_j}^{l,eq}$ of crystal phase j is computed using the polynomial function

392
$$\begin{aligned} x_{c_j}^{l,eq}(P^*, T^*, x_d^*) &= \zeta_{j,1}(P^*)^2 + \zeta_{j,2}(T^*)^2 + \zeta_{j,3}(x_d^*)^2 + \zeta_{j,4}(P^*)(T^*) + \\ &+ \zeta_{j,5}(T^*)(x_d^*) + \zeta_{j,6}(x_d^*)(P^*) + \zeta_{j,7}(P^*) + \zeta_{j,8}(T^*) + \zeta_{j,9}(x_d^*) + \zeta_{j,10}, \end{aligned}$$

393 where P^* is the liquid pressure expressed in bars, T^* is the temperature expressed in Celsius
 394 degrees and x_d^* is the dissolved water concentration in weight percent. From $x_{c_j}^{l,eq}$, the equilibrium
 395 crystal volume fraction β_j^{eq} can be computed using the relation

396
$$\beta_j^{eq} = \frac{\rho_l x_{c_j}^{l,eq}}{\rho_{c_j}}.$$

397 The parameters $\zeta_{j,i}$ are calculated fitting the polynomial function over a large range of data
 398 obtained at different pressures, temperatures and water contents with alphaMELTS⁶⁴, a command
 399 line version of MELTS⁶⁵. As previously, we used the average melt inclusion composition (Etna 122
 400 B.C.) from ref. (61).

401 Disequilibrium crystallisation is considered in our model and characteristic times controls the
 402 time needed to reach equilibrium crystal content. The differential equation in the conduit model
 403 which describes the crystal volume fraction within the conduit is the following²¹:

404
$$\frac{\partial \alpha_l \rho_c \beta u_l}{\partial z} = -\frac{1}{\tau^{(c)}} \alpha_l \rho_c (\beta - \beta^{eq}).$$

405 Here, α_l is the volume fraction of the liquid phase, ρ_c is the density of the crystals, u_l is the
 406 velocity of the liquid phase, z is the vertical axis, β is the actual total crystal volume fraction,
 407 whereas β^{eq} is the equilibrium total crystal volume fraction. If we multiply all the terms by the
 408 characteristic time $\tau^{(c)}$, we notice that the smaller is $\tau^{(c)}$, the smaller has to be $(\beta - \beta^{eq})$. In other
 409 terms, the smaller is the characteristic time, the faster the crystals will reach the equilibrium value.

410 Formally, as described in La Spina et al. (ref. 21), the characteristic time reflects the time
 411 required to reduce the difference between the actual and the equilibrium value to e^{-1} (~37%) of the
 412 initial difference. This means that, if β_0 is the initial value of a physical parameter β (which, in our
 413 case, is the crystal volume fraction), and β_{eq} is the equilibrium value in response to a perturbation of
 414 the system, at the characteristic time τ , we have

415
$$\beta(\tau) = \beta^{eq} + e^{-1}(\beta_0 - \beta^{eq}).$$

416 The experimental phase diagram for Etna basalt erupted during the 122 B.C. eruption³⁷ provides
 417 the plagioclase and pyroxene liquidus at different pressures and temperatures, whilst the conduit
 418 model is able to track temperature evolution within the conduit. Combining both of these data, we
 419 can estimate ΔT with respect to the plagioclase and pyroxene liquidus during magma ascent.

420 For this work, as we are interested in the highly explosive activity, we assumed no relative
421 velocity between gas and melt. Furthermore, as we indicated in the main text, we used as
422 fragmentation model the strain-rate criterion introduced by ref. (1).

423

424 **Initial condition for the sensitivity analysis.**

425 The range of input parameters adopted for the sensitivity analysis are the following: 140–160
426 MPa for the inlet pressure at 6000 m depth, 1050–1100 °C for the magma inlet temperature, 5–30 m
427 for the radius of the conduit, 2.0–4.0 wt.% for the total water content, 0.1–2.0 wt.% for the total
428 CO₂ content, 0–20 vol.% for the initial phenocrysts, and 1–1000 s for the characteristic time of
429 crystallisation. As we do not know the probability distribution of the uncertain input parameters,
430 we have assumed a uniform distribution within the aforementioned ranges. The sensitivity analysis
431 was performed on using the DAKOTA toolkit (Design Analysis Kit for Optimization and Terascale
432 Applications)⁶⁶, an open-source software developed at Sandia National Laboratories that provides a
433 flexible and extensible interface between analysis codes and iterative systems analysis methods
434 such as uncertainty quantification, sensitivity analysis, optimization, and parameter estimation.

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610

611

612 **Supplementary information** is available in the online version of the paper.

613

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618

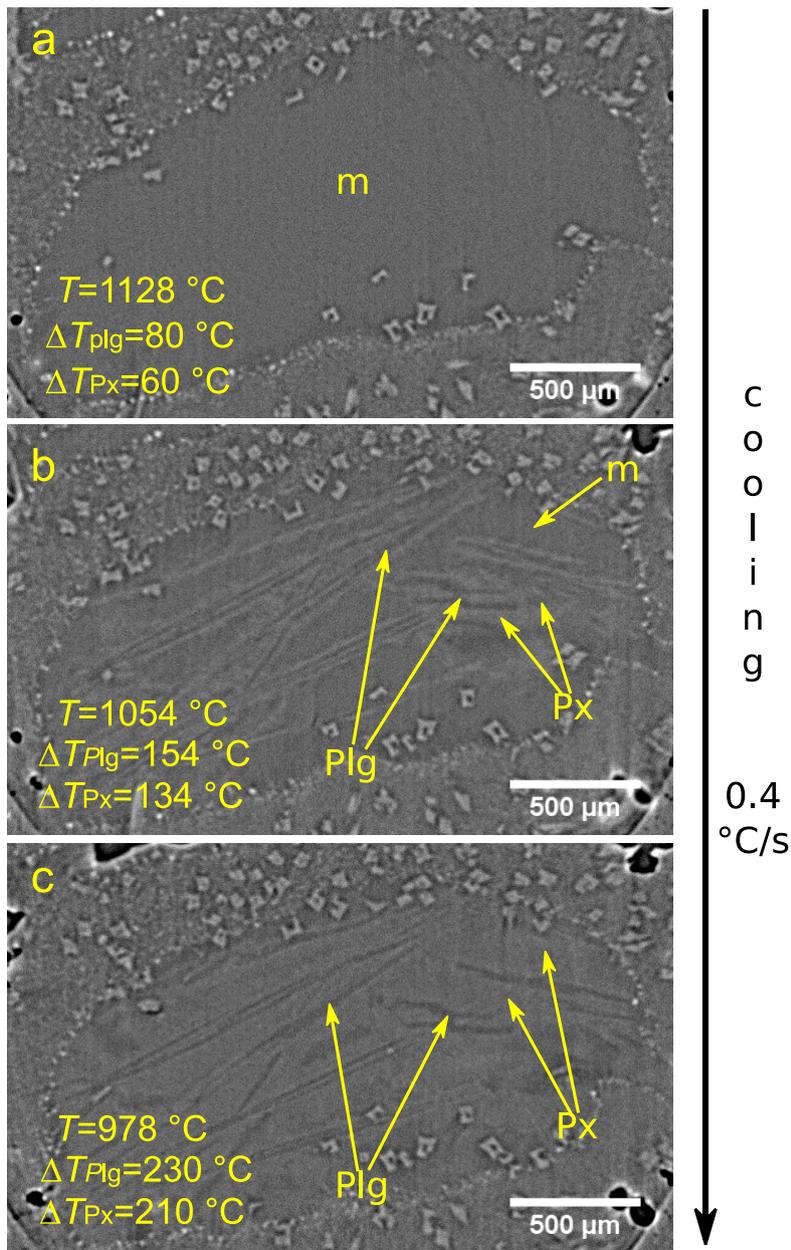
619 **Author Contributions**

620 M.P., F.A., M.R.B., and P.D.L. conceived the research project. F.A., M.P., G.L.S., N.L.G., B.C.,
621 M.E.H., D.D.G., N.T.V., S.N., R.C.A., E.W.L., P.D.L. and M.R.B. contributed to the beamline
622 experiments. F.A. collected the volcanic rocks for the starting material. D.D.G., H.M and R.A.B.
623 prepared the starting material. F.A., M.P. and G.L.S performed image reconstruction. F.A. and M.P.
624 performed image processing. F.A. performed image segmentation and analysis. G.L.S. performed
625 simulations using the conduit model. R.A.B. and F.A. performed ex situ decompression
626 experiments. E.C.B., F.A. and G.L.S. collected samples of the Etna 122 BC Plinian eruption. E.C.B.
627 and F.A. acquired and analysed back-scattered electron images of Etna Plinian eruption's samples.
628 F.A., G.L.S., M.R.B., M.P. and E.C.B. wrote the manuscript, with contributions from all other
629 authors.

630

631 **Author Information**

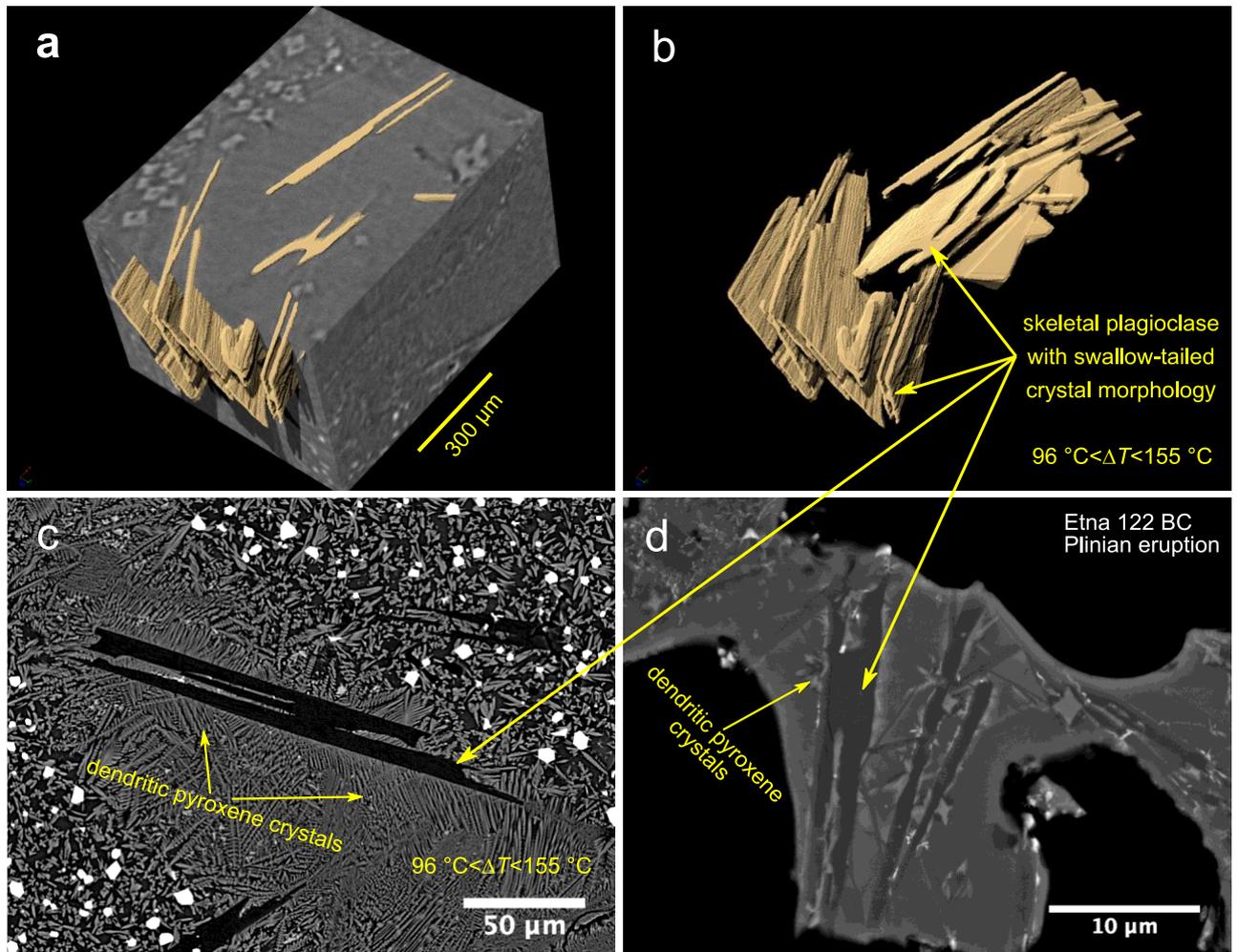
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633 declare no competing financial interests. Readers are welcome to comment on the online version of
634 the paper. Correspondence and requests for materials should be addressed to F.A.
635 (fabio.arzilli@manchester.ac.uk)



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638 Figure 1. Crystallisation through time. Reconstructed axial slices during continuous cooling at
 639 0.4 °C s^{-1} : (a) frame after 24 s from the onset of the cooling, in which the temperature ranges
 640 between 1144 and 1112 °C (average 1128 °C); (b) frame after 208 s, in which the temperature
 641 ranges between 1073 and 1034 °C (average 1054 °C); (c) frame after 392 s, in which the
 642 temperature ranges between 997 and 959 °C (average 978 °C). m = melt; plg = plagioclase; px =
 643 pyroxene.

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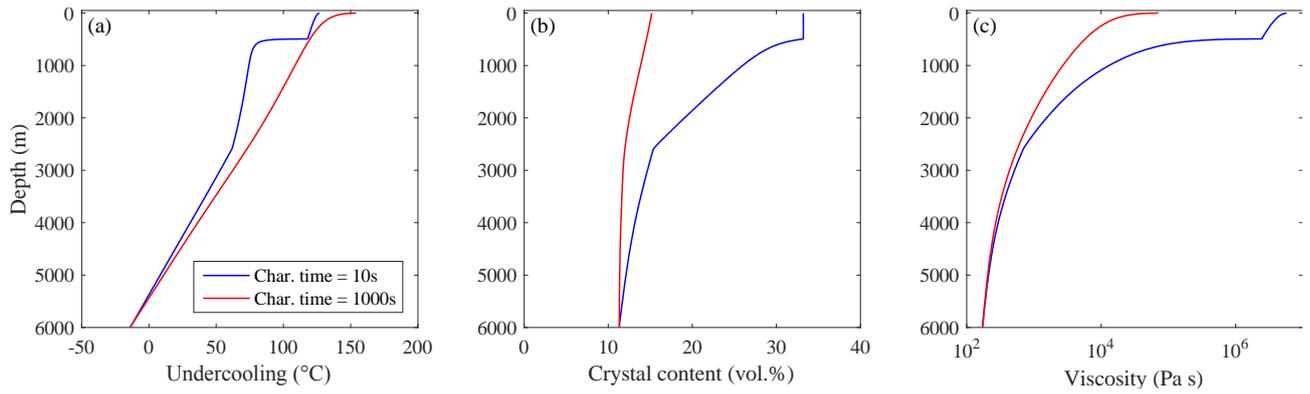


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648 Figure 2. Plagioclase crystal morphology. (a) The 3D volume rendering of sample SS1150 shows
 649 the morphology and the spatial distribution of plagioclase crystals that formed during the rapid
 650 cooling at $96 < \Delta T < 155$ °C. (b) 3D view of the plagioclase with swallow-tailed crystal morphology.
 651 (c) Back scattered electron image of plagioclase with swallow-tailed crystal morphology.
 652 scattered electron image of plagioclase with swallow-tailed crystal morphology produced during the
 653 Etna 122 B.C Plinian eruption. Note heterogeneous nucleation of pyroxene around plagioclase, seen
 654 as a light-coloured halo, and similar to that seen in figure 1b.

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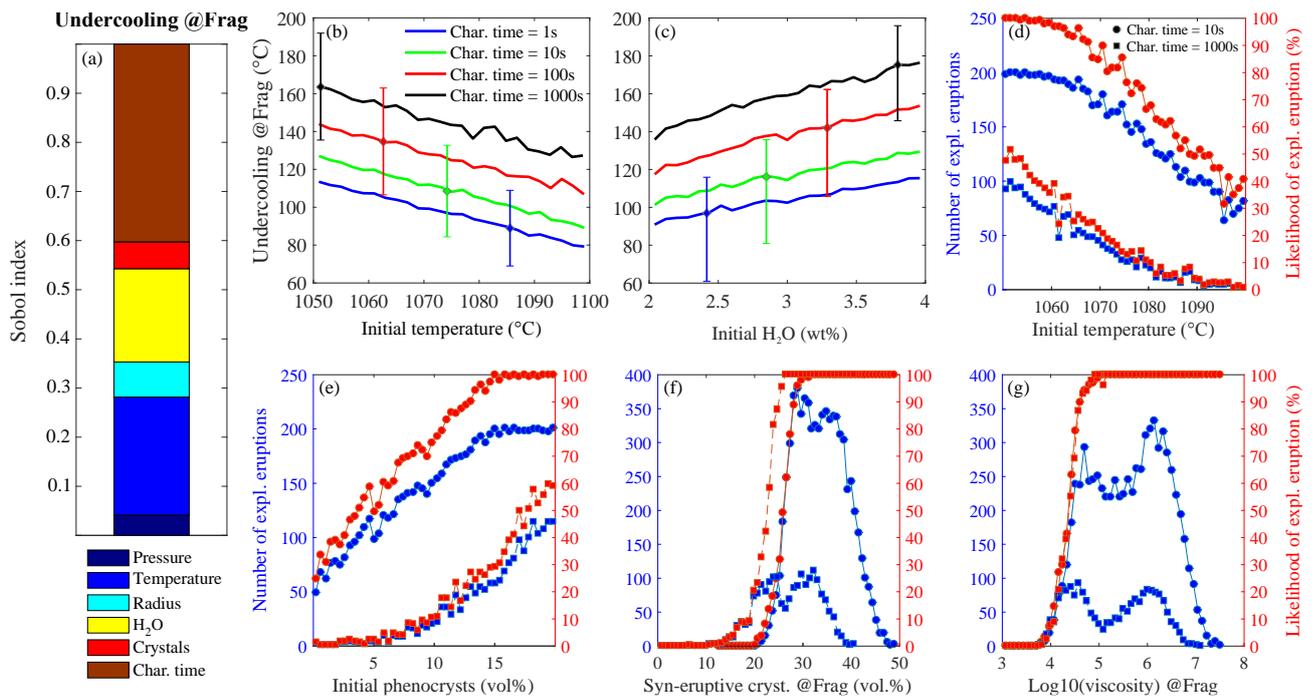


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658 Figure 3. Model results during magma ascent. (a) Undercooling as a function of depth, calculated
 659 for $\tau(c) = 10$ (blue) and 1000 s (red). Cooling is driven by adiabatic expansion of gas, mitigated by
 660 latent heat of crystallisation particularly in the fast crystallising case. (b) Crystal content in vol%,
 661 demonstrating the rapid increase in crystal load when $\tau(c) = 10$ s. (c) Magma viscosity,
 662 demonstrating that the higher crystal load produces 3-4 order of magnitude increase in viscosity,
 663 leading to fragmentation.

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667 Figure 4. Relationships between characteristic time, initial temperature, initial H₂O content of the
 668 magma, syn-eruptive crystal content and magma viscosity and the undercooling of the system at the
 669 fragmentation level. These figures were calculated using repeated runs of the model while changing
 670 individual parameters to reveal the sensitivity of the system to each parameter. Likelihood of
 671 explosive eruption as a function of a specific parameter arises from the ratio between the number of
 672 model runs producing explosive eruptions divided by the total number of model runs used to test
 673 that parameter. Therefore, this is not a probabilistic assessment of eruption risk, but instead depends
 674 on the critical model parameters, which control when fragmentation occurs, and the calculated
 675 probabilities depend on the choice of upper and lower limits chosen for each investigated
 676 parameter. (a) Sobol index. (b) Undercooling vs magma temperature before ascent. (c)
 677 Undercooling vs the initial H₂O content of the magma (dissolved and exsolved). For a given input
 678 value, the solid lines are the mean of a given output parameter obtained from all the simulations
 679 assuming that input value. The error bars, instead, represent the spread of that output values with
 680 respect to the corresponding mean. (d) Frequency of explosive eruptions vs magma temperature
 681 before ascent. (e) Explosion frequency vs initial phenocryst content. (f) Frequency of explosive

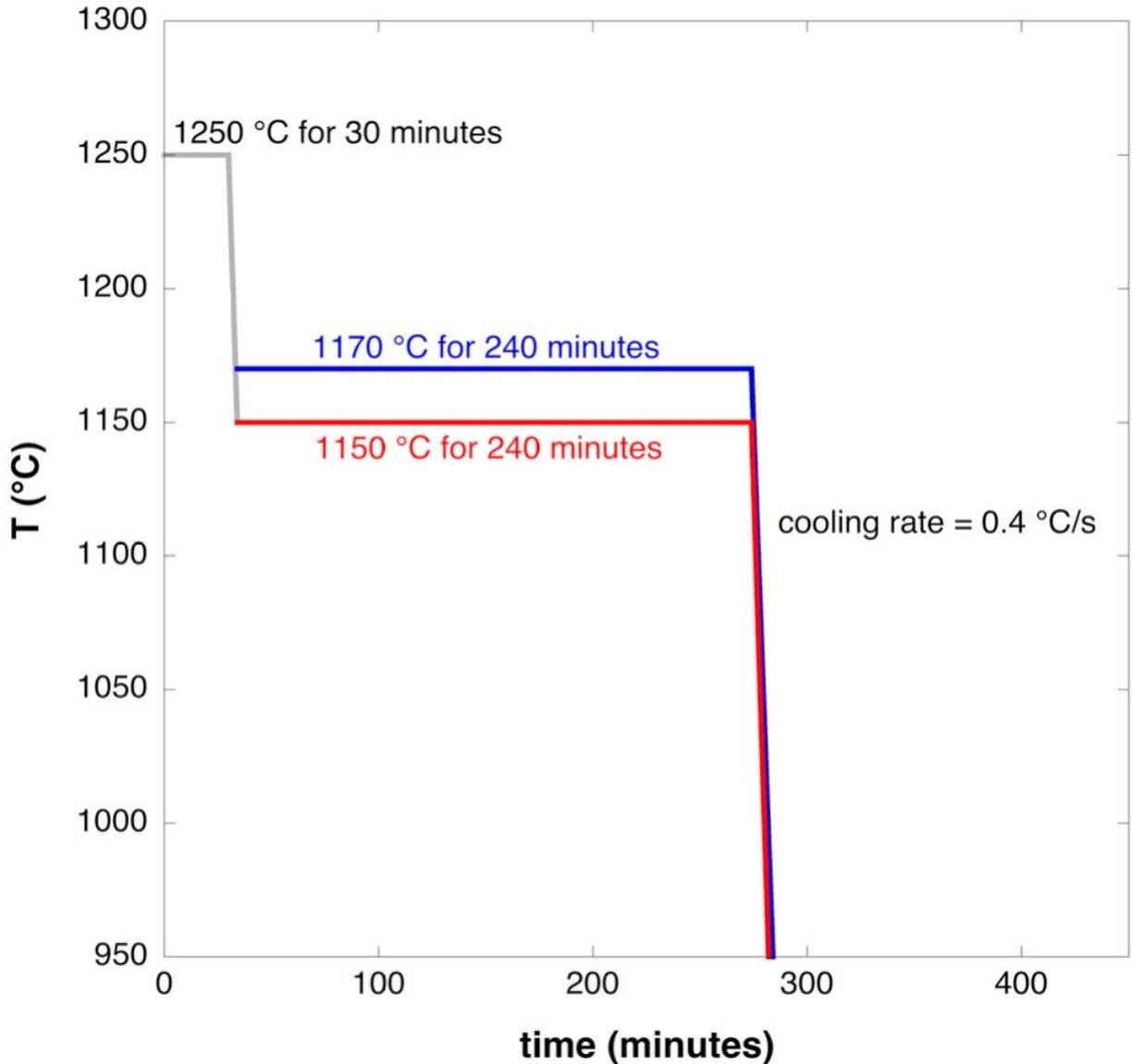
682 eruptions vs syn-eruptive crystal content at the fragmentation level. (g) Frequency of explosive

683 eruptions vs magma viscosity at the fragmentation level.

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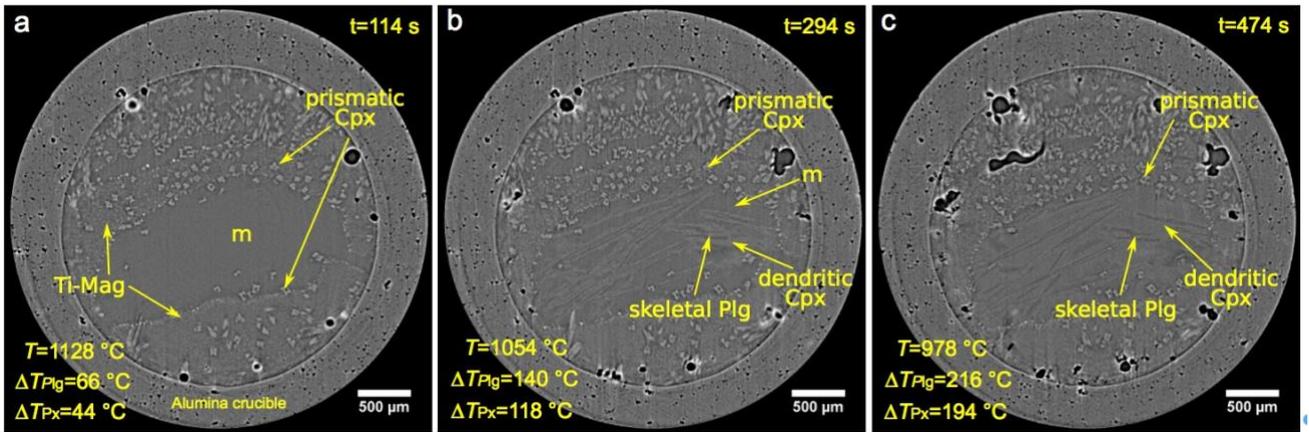
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Supplementary Information



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Supplementary Figure 1. Experimental conditions of *in situ* 4D experiments. The glassy cylinders were heated *in situ* in the Alice furnace up to 1250 °C for 30 minutes. After the initial annealing period, crystallisation was induced by decreasing temperature from 1250 °C to 1170 °C (SS1170) or 1150 °C (SS1150) at ambient pressure, holding at the final temperature for 240 minutes. After this step, the temperature was perturbed by a rapid cooling at rate of 0.4 °C/s, reaching ~950 °C in ~8 minutes.



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Supplementary Figure 2. Crystallisation through time during experiment SS1150.

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Reconstructed axial slices during continuous cooling at $0.4 \text{ } ^\circ\text{C s}^{-1}$: **a**, frame represents the texture

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after 114 s from the onset of the cooling in which the temperature ranges between 1144 and 1112

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$^\circ\text{C}$ (average $1128 \text{ } ^\circ\text{C}$); **b**, frame after 294 s, in which the temperature ranges between 1073 and 1034

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$^\circ\text{C}$ (average $1054 \text{ } ^\circ\text{C}$); **c**, frame after 474 s, in which the temperature ranges between 997 and 959

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$^\circ\text{C}$ (average $978 \text{ } ^\circ\text{C}$). m = melt; Plg = plagioclase; Cpx = clinopyroxene; Ti-Mag: titanomagnetite.

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The acquisition time of each frame is 90 s. t = time from the onset of the cooling to the end of the frame acquisition.

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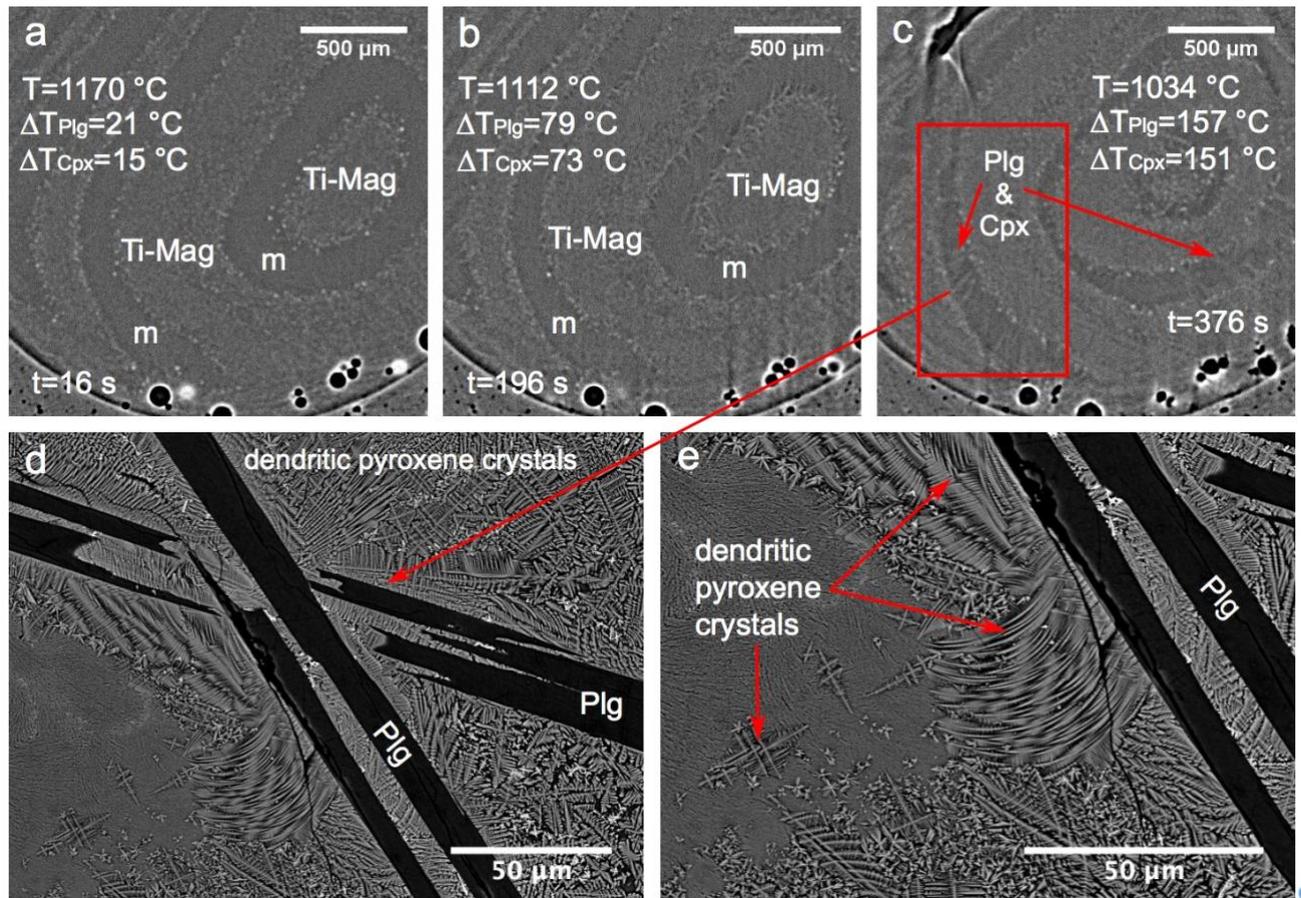
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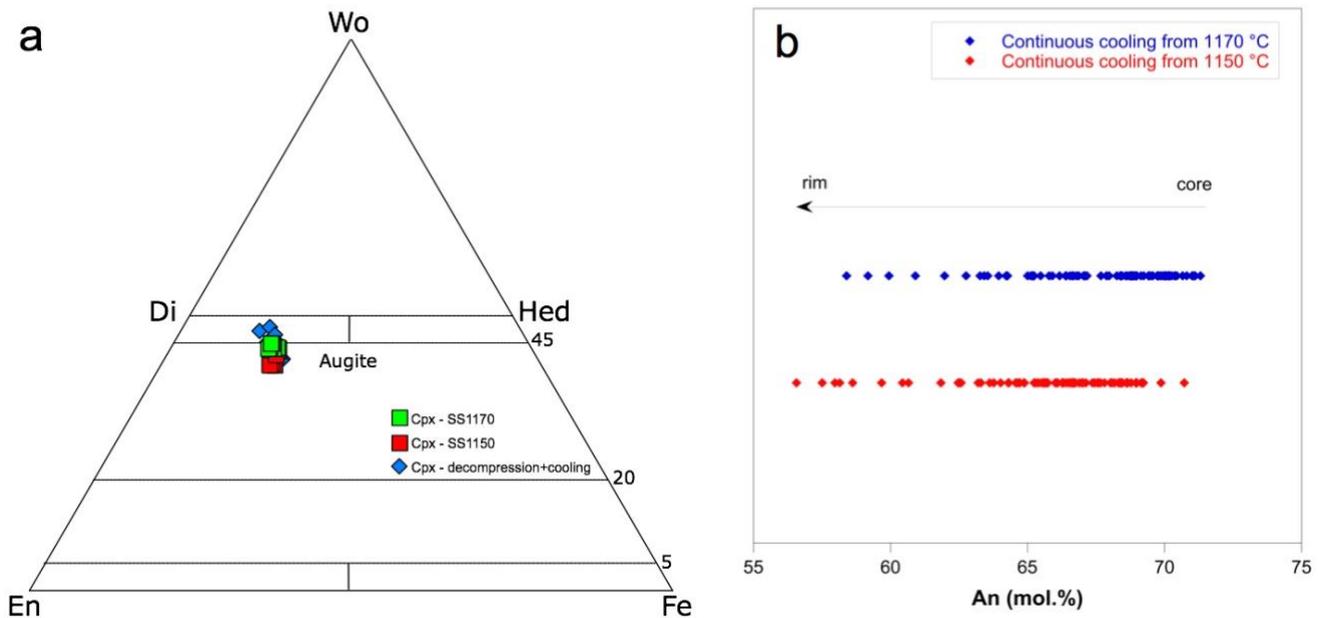
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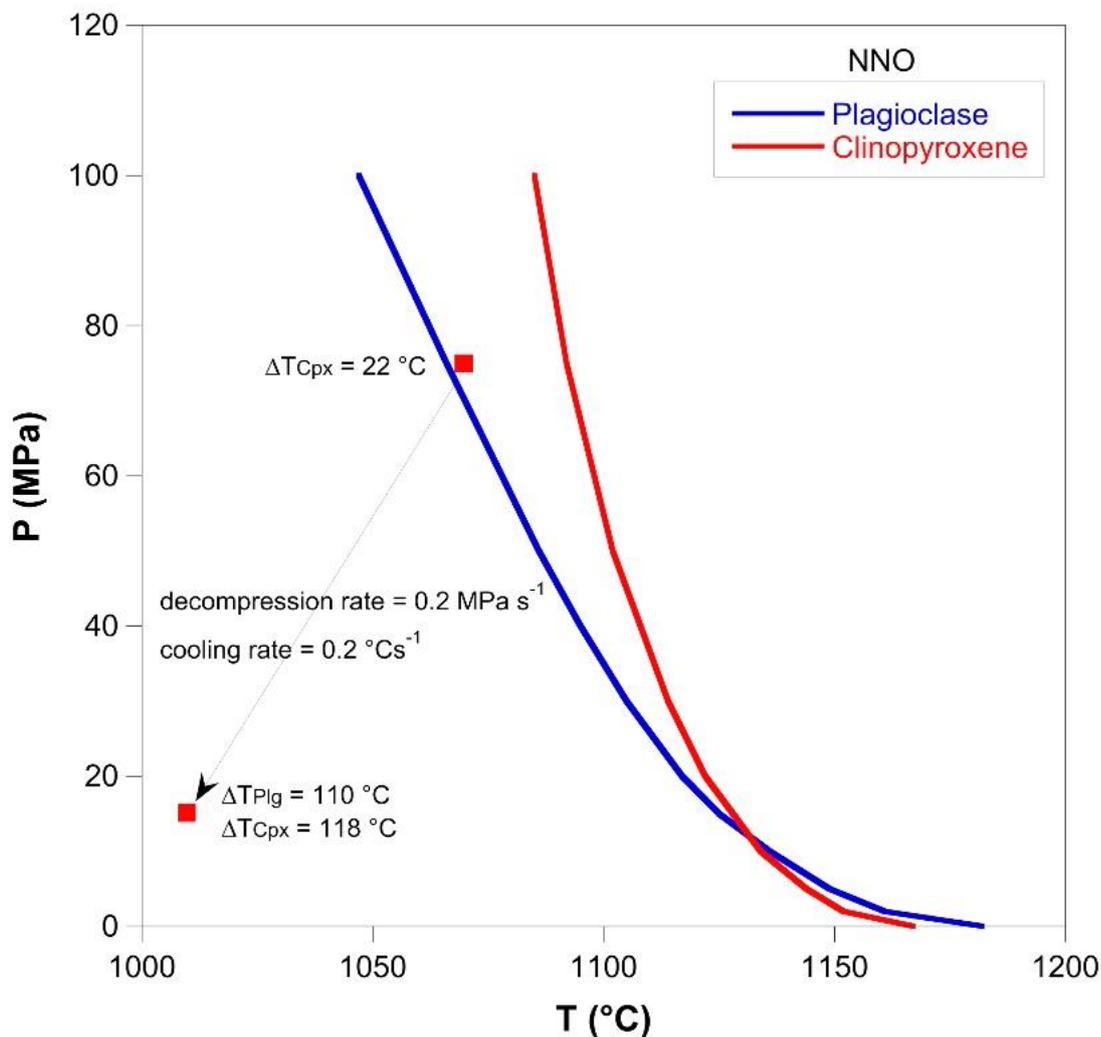
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Supplementary Figure 3. Crystallisation through time during experiment SS1170. Reconstructed axial slices during continuous cooling at $0.4\text{ }^{\circ}\text{C s}^{-1}$: **a**, this frame shows the first 16 s from the onset of the cooling, in which the temperature ranges between 1170 and 1165 $^{\circ}\text{C}$; **b**, frame after 196 s, in which the temperature ranges between 1131 and 1092 $^{\circ}\text{C}$ (average 1112 $^{\circ}\text{C}$); **c**, frame after 376 s, in which the temperature ranges between 1053 and 1015 $^{\circ}\text{C}$ (average 1034 $^{\circ}\text{C}$). m = melt; Plg = plagioclase; Cpx = pyroxene; Ti-Mag: titanomagnetite. The acquisition time of each frame is 90 s. t = time from the onset of the cooling to the end of the frame acquisition.



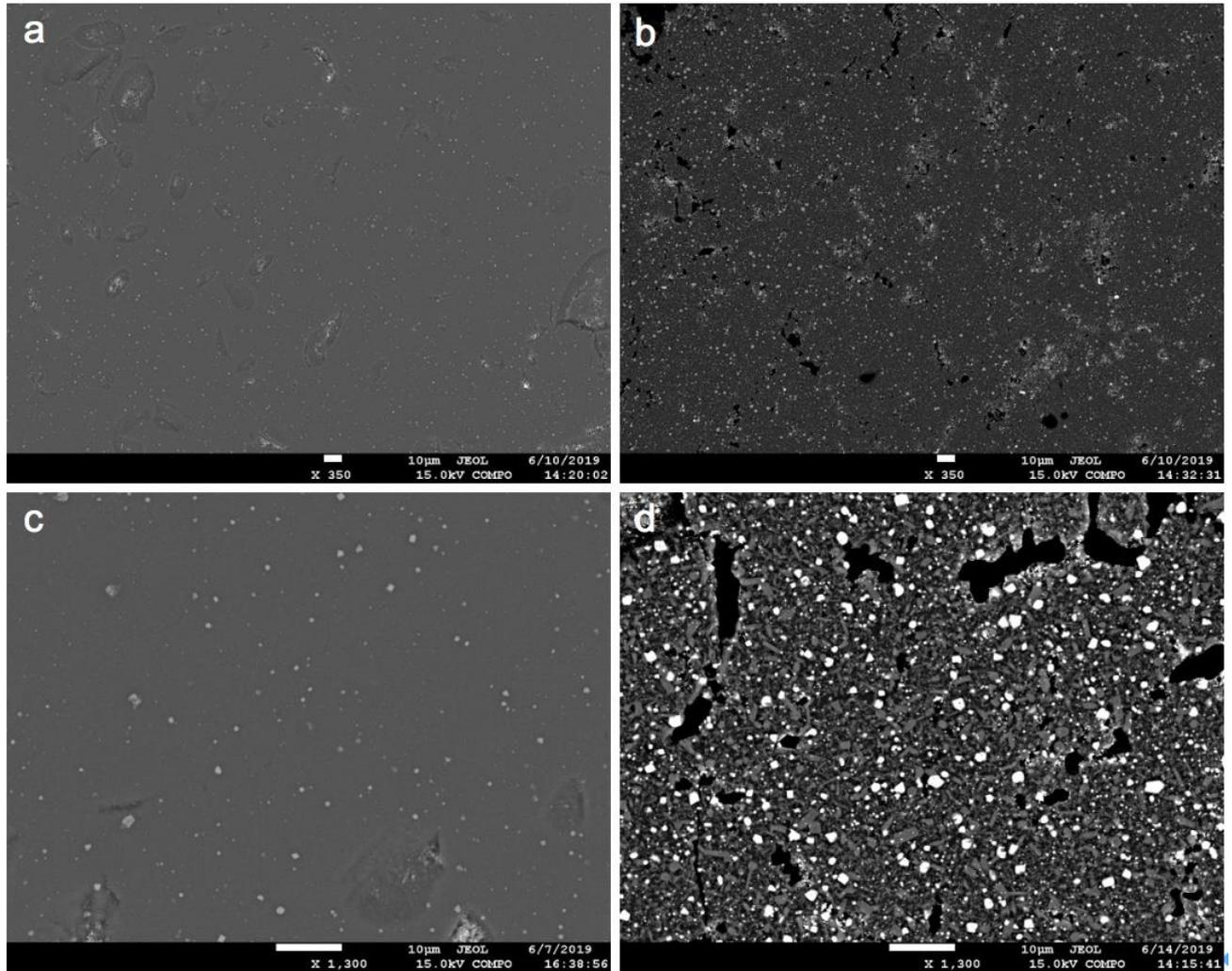
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 735 **Supplementary Figure 4. Chemical compositions of pyroxene and plagioclase formed during**
 736 **experiments. a,** Compositions of pyroxene crystals during rapid cooling at ambient pressure
 737 (SS1170 and SS1150) and rapid decompression and cooling (Etna_7_C): green square=SS1170; red
 738 square=SS1150; blue diamond=Etna_7_C. En: enstatite; Fs: ferrosilite; Di: diopside; Hd:
 739 hedenbergite; Wo: Wollastonite. **b,** Anorthite contents of plagioclase crystals formed during *in situ*
 740 experiments (SS1170 and SS1150).

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Supplementary Figure 5. *Ex situ* experimental conditions of rapid cooling and decompression experiment. The experimental approach consisted of holding the sample for 30 minutes at 75 MPa and 1070 °C before being decompressed and cooled. The first experiment was quenched after 30 minutes at 75 MPa and 1070 °C, in order to texturally characterise the initial conditions before that fast decompression and cooling were applied. In the second experiment, after 30 minutes pressure and temperature were decreased continuously in 300 s at 0.2 MPa s⁻¹ and 0.2 °Cs⁻¹ respectively, reaching 15 MPa and 1010 °C. Final pressure was chosen as approximate fragmentation pressures calculated by modeling. Decompression and cooling rates were predicted from the conduit model used in this study. Plagioclase and clinopyroxene liquids were obtained with Rhyolite-MELTS software (version 1.2), using the composition of our starting material.



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 767 **Supplementary Figure 6.** Textures of *ex situ* experiments. Back scattered images show the textures of:
 768 **a, c**, an hydrous basalt quenched at 75 Mpa and 1070 °C after 30 minutes (sample Etna7_C) and
 769 **b, d**, an hydrous basalt quenched after a rapid decompression and cooling at 15 MPa and 1010 °C
 770 (sample Etna7_B). **a, c**, The texture of sample Etna7_C is characterised by oxide crystals and glassy
 771 matrix. **b, d**, The mineral assemblage of sample Etna7_B consists of oxide, clinopyroxene and
 772 plagioclase. Oxide formed mostly at initial condntions, after 30 minutes at 1070 °C and 75 MPa (see
 773 Etna7_C, panels **a** and **c**). The rapid decompression and cooling allowed the achievement of large
 774 ΔT s (>100 °C; see Supplementary Fig. 6) in 300 s, favouring predominantly clinopyroxene
 775 crystallisation, plagioclase and oxide. This indicates that $\tau^{(c)}$ is 60 s, in agreement with the order of
 776 magnitude estimated from *in situ* experiments.

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789 **Supplementary Table 1.** Chemical composition of the starting material.

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Oxide (wt.%)	Bulk composition	st. dev.	Residual melt after 4h at 1170 °C	Residual melt after 4h at 1150 °C
SiO ₂	48.41	0.16	52.18	52.99
TiO ₂	1.79	0.07	1.7	1.69
Al ₂ O ₃	16.22	0.50	16.86	17.5
FeO*	10.66	0.30	6.05	5.4
MnO	0.20	0.03	0.22	0.24
MgO	6.20	0.32	5.34	4.77
CaO	10.66	0.19	11.23	10.54
Na ₂ O	3.41	0.20	3.74	3.99
K ₂ O	1.90	0.07	2.09	2.24
P ₂ O ₅	0.54	0.04	0.59	0.64
Total	99.85		100	100

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note: FeO* = total iron as FeO. The residual melt compositions after the 240 minutes have been obtained with a mass balance calculation, which indicates that they are basaltic trachyandesite. We considered that the main difference between experiment SS1150 and SS1170 is that the abundance of pyroxene after 4 hours at 1150 °C is higher than that at 1170 °C. Experiment SS1170 (after 4 hour at 1170 °C) indicates that prismatic pyroxene crystals are ~ 1 vol.% and oxide crystals are ~8 vol. %, whereas in the experiment SS1150 (after 4 hour at 1150 °C) prismatic pyroxene crystals are more abundant (~8 vol.%).

Supplementary Table 2. Results of plagioclase and pyroxene growth rates during continuous cooling at 0.4 °C/s.

3D analysis							
<i>Plagioclase crystal growth rate</i>							
Sample	Growth duration (s)	L3Dmax (cm)	YL3Dmax (cm s ⁻¹)	L3Dmin (cm)	YL3Dmin (cm s ⁻¹)	L3Dave (cm)	YL3Dave (cm s ⁻¹)
SS1150	90	2.77E-02	1.54E-04	6.35E-03	3.53E-05	1.39E-02	7.74E-05
2D analysis							
<i>Plagioclase crystal growth rate</i>							
Sample	Growth duration (s)	Lmax (cm)	YLmax (cm s ⁻¹)	Lmin (cm)	YLmin (cm s ⁻¹)	Lave (cm)	YLave (cm s ⁻¹)
SS1170	180	2.73E-02	7.59E-05	1.15E-02	3.20E-05	1.78E-02	4.96E-05
<i>Pyroxene crystal growth rate</i>							
Sample	Growth duration (s)	Lmax (cm)	YLmax (cm s ⁻¹)				
SS1170	180	3.6E-03	2.0E-05				
SS1150	180	3.24E-03	1.80E-05				

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note: L3Dmax = average length of the 15 largest crystals measured in the 3D domain; YL3Dmax = growth rate calculated using the 15 largest crystals; L3Dmin = average length of the 15 smallest crystals measured in the 3D domain; YL3Dmin = growth rate calculated using the 15 smallest crystals; YL3Dave = the average value of growth rate obtained taking into account the length all the crystals analysed; Lmax = average length of the 15 largest crystals measured in the 2D domain; YLmax = growth rate calculated using the 15 largest crystals.

812 **Supplementary Table 3.** Experimental conditions for ex situ experiments performed with TZM
 813 cold seal pressure vessel apparatus.

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Sample	Ti (°C)	Pi (Mpa)	ti (s)	decompression rate (Mpa s-1)	cooling rate (°Cs-1)	Tf	Pf
Etna7_C	1070	75	1800	-	-	1070	75
Etna7_B	1070	75	1800	0.2	0.2	1010	15

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820 **Supplementary Table 4.** Summary of X-ray microtomography acquisition conditions and image
 821 processing details of the volumes of interest (VOI) for experiment ET1150.

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Energy	Distance sample to detector	Projection	Exposure time	Voxel size	Original imaged volume	Analysed VOI	Segmentation
keV	mm	number	s	μm^3	pixels (mm^3)	mm^3	
53	2300	1800	0.05	3.2x3.2x3.2	1683x1683x1383 (128.4)	8.2037	semi-automatic volume segmentation Avizo software v. 8.0

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