



Maury, J., Zawodny, W., & Clayden, J. (2017). Stereospecific Intramolecular Arylation of 2- and 3-Pyridyl Substituted Alkylamines via Configurationally Stable  $\alpha$ -Pyridyl Organolithiums. *Organic Letters*, 19(3), 472-475. <https://doi.org/10.1021/acs.orglett.6b03603>

Peer reviewed version

Link to published version (if available):  
[10.1021/acs.orglett.6b03603](https://doi.org/10.1021/acs.orglett.6b03603)

[Link to publication record on the Bristol Research Portal](#)  
PDF-document

This is the author accepted manuscript (AAM). The final published version (version of record) is available online via ACS at <http://pubs.acs.org/doi/abs/10.1021/acs.orglett.6b03603>. Please refer to any applicable terms of use of the publisher.

## University of Bristol – Bristol Research Portal

### General rights

This document is made available in accordance with publisher policies. Please cite only the published version using the reference above. Full terms of use are available: <http://www.bristol.ac.uk/red/research-policy/pure/user-guides/brp-terms/>

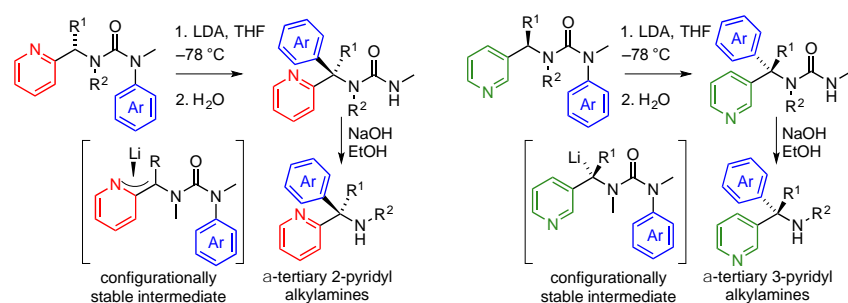
# Stereospecific Intramolecular Arylation of 2- and 3-Pyridyl Substituted Alkylamines via Configurationally Stable $\alpha$ -Pyridyl Organolithiums

Julien Maury<sup>‡,a,b</sup>, Wojciech Zawodny<sup>‡,a</sup> and Jonathan Clayden<sup>b\*</sup>

<sup>a</sup> School of Chemistry, University of Manchester, Oxford Road, Manchester M13 9PL, U.K.

<sup>b</sup> School of Chemistry, University of Bristol, Cantock's Close, Bristol BS8 1TS, U.K.

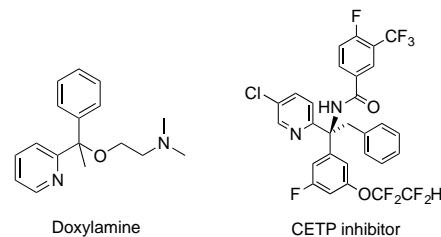
Supporting Information Placeholder



**ABSTRACT:** Treatment of  $N'$ -aryl urea derivatives of enantiomerically-enriched  $\alpha$ -(2-pyridyl) and  $\alpha$ -(3-pyridyl)alkylamines with base leads to the migration of the  $N'$ -aryl substituent from  $N$  to  $C$  in a 'non-classical' intramolecular nucleophilic aromatic substitution reaction. Both electron-rich and electron-poor rings migrate successfully. A new quaternary stereogenic centre is formed adjacent to the pyridine ring with high stereospecificity, even when the intermediate anion is a presumably planar 2-picolylolithium. Base hydrolysis of the urea gives enantiomerically enriched  $\alpha$ -pyridylalkylamines.

Pyridines play a vital role in medicinal chemistry,<sup>1</sup> being the most common heterocyclic ring encountered in small molecule drugs.<sup>2</sup> Substituted chiral pyridines with a stereogenic centre at the 'picolinic' position  $\alpha$  to the pyridine ring are present in many biologically active molecules<sup>3</sup> and chiral ligands.<sup>4</sup> More specifically, congested quaternary stereogenic centres bearing both pyridine and phenyl rings are present in antihistamines such as pheniramine<sup>5</sup> and doxylamine<sup>6</sup> and in potent cholesteryl ester transfer protein (CETP) inhibitors (Figure 1.).<sup>7</sup>

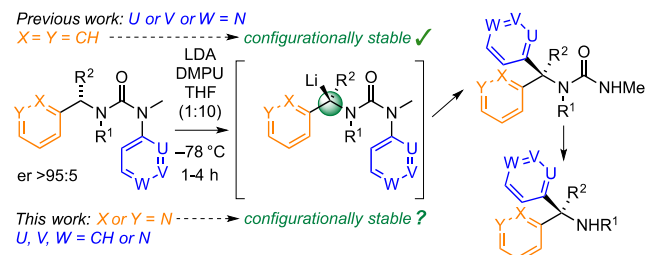
Figure 1. Bioactive compounds with a quaternary stereogenic centre  $\alpha$  to a pyridine ring



Methods for the synthesis of  $\alpha$ -chiral amines bearing a pyridine ring at the stereogenic centre typically rely on auxiliaries to direct addition to, or reduction of, an intermediate imine.<sup>8</sup> For enantiopure  $\alpha$ -tertiary amines with a pyridine ring as one of the substituents at the quaternary

stereogenic centre (Figure 1) synthetic approaches are very limited.<sup>9</sup> We previously reported a stereospecific route to a subclass of these structures by intramolecular migration of pyridine rings to the  $\alpha$ -position of lithiated urea derivatives of  $N$ - $\alpha$ -methylbenzylamine, giving  $\alpha$ -tertiary amines after the solvolysis of the urea (Scheme 1).<sup>9c</sup> Stereospecificity is ensured by the configurational stability of the benzylolithium intermediates<sup>10</sup> on the time scale of the rearrangement reaction.<sup>11</sup>

Scheme 1.  $\alpha$ -Pyridylation of chiral amines.<sup>2</sup>

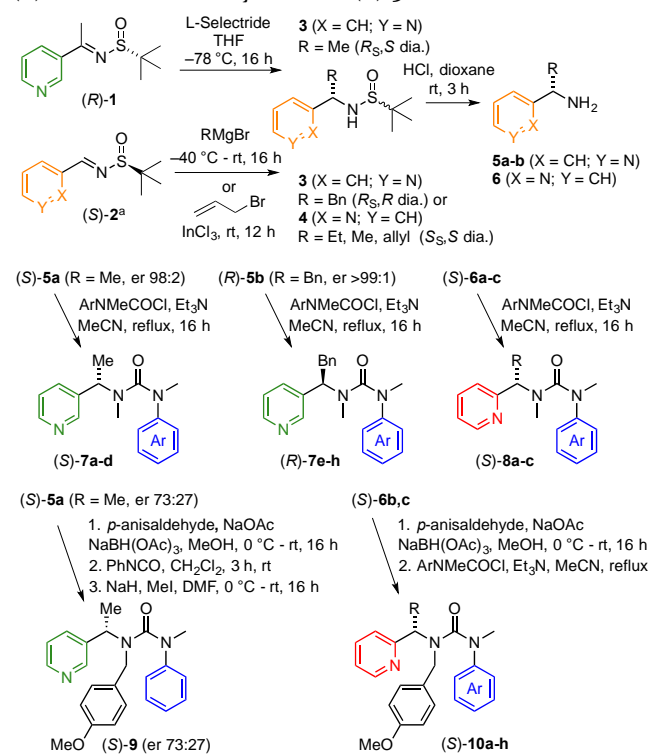


We now report a complementary method for the generation of pyridine-bearing quaternary stereogenic centers by stereospecific intramolecular arylation<sup>12</sup>  $\alpha$  to 2- or 3-pyridyl substituents. The reaction is mediated by pyridine-stabilised organolithiums that exhibit remarkable

configurational stability, given the electron-withdrawing, anion-stabilising nature of the pyridyl substituents.

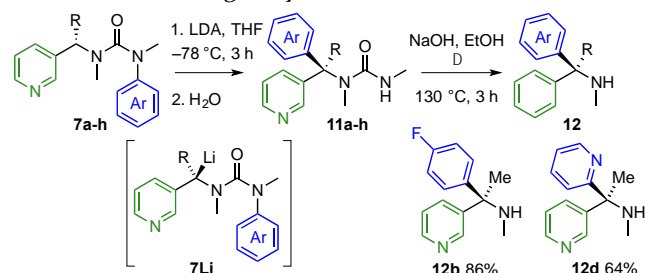
Chiral amine precursors **5** and **6** bearing an amino group and a 3- or 2-pyridyl substituent at the stereogenic centre were made using Ellman's *N*-sulfinyl auxiliary.<sup>8</sup> Grignard addition<sup>13</sup> and indium allylation<sup>14</sup> of *N*-sulfinyl aldimines **2** or diastereoselective reduction of *N*-sulfinyl ketimines<sup>15</sup> **1** gave highly enantioenriched sulfinamides **3** and **4** that were hydrolysed to chiral primary amines **5** and **6** (Scheme 2). The amines were either acylated (with *N*-methylcarbamoyl chlorides) and methylated, or reductively aminated with *p*-methoxybenzaldehyde and acylated (with aryl isocyanates, followed by methylation, or with *N*-methylcarbamoyl chlorides), giving ureas **7-10** as starting materials for organolithium-mediated rearrangements.

Scheme 2. Synthesis of enantioenriched 2- and 3-pyridyl-substituted urea starting materials. <sup>a</sup>Enantiomeric sulfinimine (*R*)-**2** was used for the synthesis of (*R*)-**5**.



The 3-pyridyl-substituted urea **7a** (R = Me, Ar = Ph) was treated with LDA in THF at -78 °C (Scheme 3). After 3 hours, the reaction was quenched, and rearranged urea **11a**, in which the phenyl group had migrated from nitrogen to the position  $\alpha$  to the pyridyl ring, was isolated in good yield without loss of enantiomeric purity (Table 1, entry 1). No additives<sup>16</sup> were required to maintain the stereospecificity<sup>17</sup> of the reaction, indicating that the presumed intermediate 3-pyridine-stabilised organolithium **7Li** does not racemise on the time scale of the rearrangement. The electron-deficient *para*-fluorophenyl, *para*-chlorophenyl, and 2-pyridyl rings of **7b-d** likewise migrated to give **11b-d** in good yields, again with full stereospecificity (entries 2-4). Urea **11d**, formed in 98:2 er, provides the first example of an  $\alpha$ -tertiary amine derivative with both a 2- and a 3-pyridyl substituent at the stereogenic center. The tolerance of the reaction to steric

hindrance<sup>18</sup> was explored by replacing the  $\alpha$ -methyl substituent with a benzyl group to build structural analogues of the CETP inhibitor in Figure 1 (entries 5-8). The rearrangements of **7e-7h** were fully stereospecific, and **11e** and **11h** were formed in good yields.



Scheme 3. Stereospecific intramolecular arylation of 3-pyridine-substituted ureas. For simplicity, reactions of the *S* enantiomers are shown; in the case of **7e-h** the *R* enantiomer was used (see Table 1).

Table 1. Arylation of 3-pyridine-substituted ureas

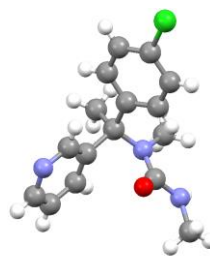
entry	SM, er <sup>a</sup>	R	Ar	product, yield (%)	product, er <sup>a</sup>
1	( <i>S</i> )- <b>7a</b> 98:2	Me	C <sub>6</sub> H <sub>5</sub>	( <i>R</i> )- <b>11a</b> 65	98:2
2	( <i>S</i> )- <b>7b</b> 98:2	Me	4-FC <sub>6</sub> H <sub>4</sub>	( <i>R</i> )- <b>11b</b> 74	98:2
3	( <i>S</i> )- <b>7c</b> 98:2	Me	4-ClC <sub>6</sub> H <sub>4</sub>	( <i>R</i> )- <b>11c</b> 63	98:2
4	( <i>S</i> )- <b>7d</b> 98:2	Me	2-pyridine	( <i>S</i> )- <b>11d</b> 76	98:2
5	( <i>R</i> )- <b>7e</b> >99:1	Bn	C <sub>6</sub> H <sub>5</sub>	( <i>S</i> )- <b>11e</b> 87	>99:1
6	( <i>R</i> )- <b>7f</b> >99:1	Bn	3-ClC <sub>6</sub> H <sub>4</sub>	( <i>S</i> )- <b>11f</b> 22	>99:1
7	( <i>R</i> )- <b>7g</b> >99:1	Bn	<sup>3</sup> -MeOC <sub>6</sub> H <sub>4</sub>	( <i>S</i> )- <b>11g</b> 26	>99:1
8	( <i>R</i> )- <b>7h</b> >99:1	Bn	2-pyridine	( <i>R</i> )- <b>11h</b> 88	>99:1

<sup>a</sup> Enantiomeric ratio by HPLC on chiral stationary phase.

Hydrolysis of the rearranged products under basic conditions cleaved the urea in good yield to provide  $\alpha$ -tertiary amines **12**, illustrated by the formation of **12b** and **12d** (Scheme 3).

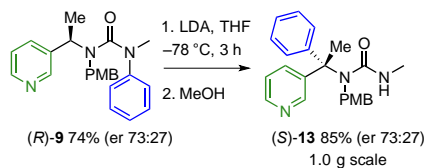
The absolute configuration of (*R*)-**11c** was determined by means of X-ray crystallography (Figure 2), and confirmed that the rearrangements of 3-pyridyl ureas proceed with retention of configuration, as has been observed in previously related rearrangements<sup>11</sup>.

Figure 2. X-ray crystal structure of (*R*)-**11c**.

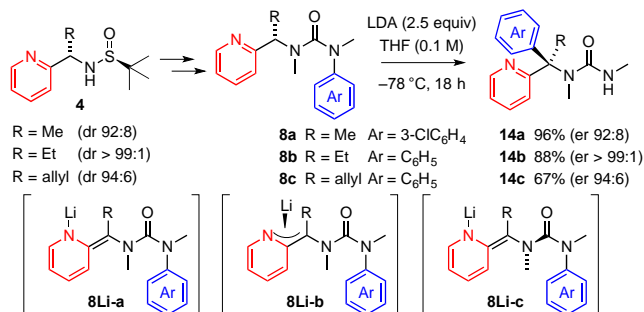


Replacing the *N*-methyl substituent with an *N*-*p*-methoxybenzyl (PMB) protecting group had no effect on the stereospecificity of the reaction: samples of (*R*)-**9** rearranged to urea **13** in good yield without loss of er on both 200 mg and 1 g scales (Scheme 4).

Scheme 4. Stereospecific intramolecular arylation of an amine with a PMB (*p*-methoxybenzyl) protecting group.



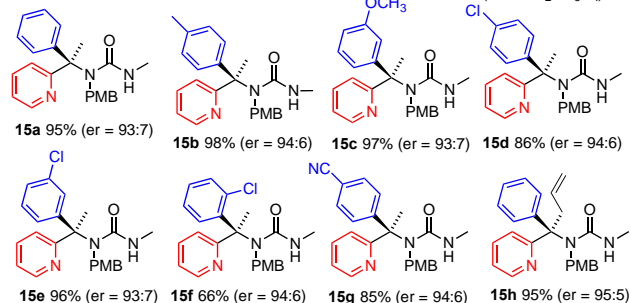
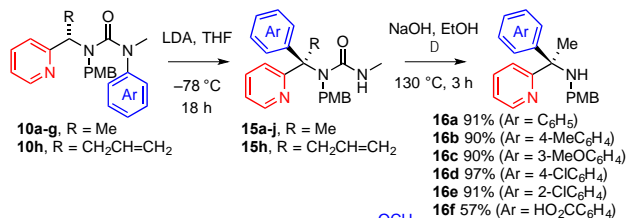
Scheme 5. Stereospecific aryl migration within 2-pyridyl-stabilized anions.



2-Pyridyl-substituted organolithiums (2-picolylolithiums) have structures best characterized as azaenolates, as represented in Scheme 5 as **8Li-a**, with a planar  $\alpha$ -carbon and the negative charge located principally at nitrogen.<sup>19</sup> Urea-substituted enolates possessing other stereogenic centres undergo diastereoselective intramolecular arylation,<sup>20</sup> but (except for examples with special structural features promoting chiral memory by hindered rotation<sup>21</sup>) without stereospecificity at the planar enolate carbon atom.<sup>22</sup> Nonetheless, treatment of the 2-pyridylurea **8a** with LDA in THF gave the rearranged compound **14a** with the same er (92:8) as the starting material (Scheme 5). Similarly, there was no loss of er in the rearrangements of ureas **8b** and **8c** bearing ethyl and allyl groups at the stereogenic centre: product ureas **14b** and **14c** were obtained in good yield and er (88%, 99:1 er and 67%, 94:6 er).

The stereospecificity of the  $\alpha$ -(2-pyridyl)alkylamine synthesis was exploited by rearrangement of a range of *p*-methoxybenzyl-protected ureas **10a-h** built from chiral 2-pyridylamines (Scheme 6) to products **15a-h** without erosion of enantiomeric enrichment. Migrating rings with either electron-donating or withdrawing substituents at the *para* and *meta* positions all rearranged in high yield (85–98%) and good er (94:6–93:7) (**15b-15e**, **15g**). The migration of aromatic groups substituted at the *ortho* position gave lower yields: with 2-chlorophenylurea **10f** the reaction remained stereospecific but the yield dropped to 66%. Attempted rearrangement of a 1-naphthyl-substituent failed. Hydrolysis of the ureas **15** under basic conditions (NaOH, EtOH) gave the valuable 2-pyridyl substituted  $\alpha$ -tertiary amines **16** in high yields (Scheme 6).<sup>23</sup>

Scheme 6. Synthesis of protected tertiary  $\alpha$ -(2-pyridyl)benzylamines by stereospecific intramolecular arylation. PMB = *p*-methoxybenzyl



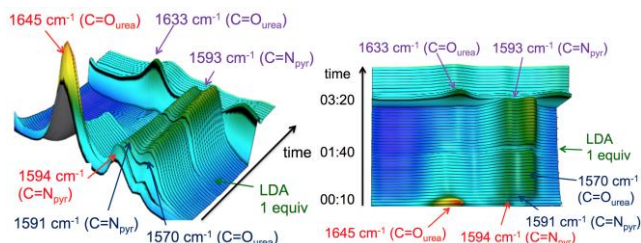
We assume that in the rearrangements of ureas **7-10** the reaction mechanism proceeds by selective deprotonation of the acidic 'picolinic' position  $\alpha$  to the pyridyl ring to give an anionic species **7-10Li**, which undergoes the conformational reorganisation required to attack the aryl ring borne by the other nitrogen atom, but without loss of configurational integrity. A non-classical intramolecular S<sub>N</sub>Ar reaction<sup>1b</sup> leads to the product anion, which is protonated on work-up. The intermediate anionic species **7-10Li** must retain their absolute stereochemistry on the time scale of the rearrangement. This stereospecificity is a feature of the reaction shared with other related rearrangements of lithiated ureas<sup>9c,11a,24</sup> (along with thiocarbamates<sup>25</sup> and, to a lesser extent, carbamates<sup>16,26</sup>), though not the cyclic ureas so far explored.<sup>27</sup>

The structure of 3-picolylolithiums related to **7-8Li** have not been examined in detail, but the inability of the nitrogen atom at the 3-position to stabilize the negative charge by delocalization suggests they may have structural similarities with configurationally stable  $\alpha$ -nitrogen substituted benzylolithiums.<sup>3</sup> By contrast, computational and crystallographic studies of 2-picolylolithiums related to **9-10Li**, whose negative charge is stabilized by delocalization onto the pyridyl nitrogen, show that the negative charge is principally located at the nitrogen atom, and the anion may be interpreted as a planar azaenolate, i.e. **8Li-a** (Scheme 5).<sup>19</sup> Given the probable planarity of the  $\alpha$ -carbon of the intermediate anion **8Li**, possible mechanisms for stereospecificity include long-lived planar chirality within an intermediate pyridyllithium species **8Li-b** (Scheme 5)<sup>25b-c,28</sup> or the adoption by the urea of a chiral, twisted conformation (such as **8Li-c**) that rearranges to product faster than it can relax to an enantiomeric mixture of conformers.<sup>21,29</sup> Attempts to probe the role of Li in the stereospecificity were frustrated by our inability to induce rearrangement with other bases (e.g. KHMDS).

Figure 3. In situ infra-red study of the rearrangement of **8b** to **11b**. 00:10:00 – Addition of 2.5 equiv LDA starts; 01:40:00 – Addition of 1 equiv LDA complete; 03:30:00 – Reaction quenched with MeOH.

(a) (b)





Scheme 7. Proposed mechanistic pathway from **8b** to **14b**

To gain deeper insight into the mechanism of the reaction, the conversion of **8b** to **14b** in THF at  $-78\text{ }^{\circ}\text{C}$  was followed by *in situ* infra-red spectroscopy (React-IR) (Figure 3 and Scheme 7). In THF at  $-78\text{ }^{\circ}\text{C}$ , IR shows one C=O stretching absorption at  $1645\text{ cm}^{-1}$  and one pyridine C=N stretching absorption at  $1594\text{ cm}^{-1}$  (Figure 3a). After 10 min (00:10) an initial 2.5 equiv LDA was added, causing the C=O stretch at  $1645\text{ cm}^{-1}$  to diminish, the pyridine stretch to shift to  $1591\text{ cm}^{-1}$ , and a new C=O stretch to grow at  $1570\text{ cm}^{-1}$ . We assign these peaks to the rearranged, lithiated urea **C**.<sup>11a,30</sup> Adding another equivalent of LDA 90 minutes later (01:40) completes the reaction, as indicated by the disappearance of the C=O stretch (starting material **A**) at  $1645\text{ cm}^{-1}$  and a further increase of the C=O stretch (lithiated product **C**) at  $1570\text{ cm}^{-1}$ . The detailed mechanism of formation of **C** from **B** was not explored, but previous studies<sup>11b</sup> have suggested that related reactions proceed by a partially concerted  $\text{S}_{\text{N}}\text{Ar}$  reaction. Finally (Figure 3b) addition of MeOH (03:20) protonates the urea anion of **C** to give **14b**, with a urea C=O stretch at  $1633\text{ cm}^{-1}$ . No species identifiable as lithiated starting material **A** was observed. Product **14b** was recovered in 89% yield.

In summary, both  $\alpha$ -(2-pyridyl) and  $\alpha$ -(3-pyridyl) alkylamines may be arylated with total enantiospecificity at their  $\alpha$ -position by intramolecular migration of an aryl substituent within their lithiated  $N^1$ -aryl urea derivatives. Despite their delocalized structure, the intermediate 2-pyridyl-substituted anions are configurational stable on the time scale of the rearrangement.

## ASSOCIATED CONTENT

### Supporting Information

Full details of experimental procedures, characterization data and spectra of all new compounds; CIF file of (*R*)-**11c**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## AUTHOR INFORMATION

‡These authors contributed equally to this work. \* Corresponding author e-mail: [j.clayden@bristol.ac.uk](mailto:j.clayden@bristol.ac.uk).

## ACKNOWLEDGMENT

This work was supported by the Marie Curie Intra-European Fellowship (IEF) “ENOLAR” and by the BBSRC. We thank Dr James Raftery (University of Manchester) for determining the X-ray crystal structure of (*R*)-**11c**.

## REFERENCES

- (a) Roughley, S. D.; Jordan, A. M.; *J. Med. Chem.*, **2011**, *54*, 3451. (b) Carey, J. S.; Laffan, D.; Thomson, C.; Williams, M. T.; (c) Vitaku, E.; Smith, D. T.; Njardarson, J. T. *J. Med. Chem.* **2014**, *57*, 10257. (d) Yadav, A.; Verma, A.; Patel, S.; Kumar, A.; Rathore, V.; Meenakshi; Kumar, Sh.; Kumar, Sa. *Chem. Commun.* **2015**, *51*, 11658. (e) Goetz, A. E.; Garg, N. K. *Nature Chem.* **2013**, *5*, 54. (f) Neely, J. M.; Rovis, T. *Org. Chem. Front.* **2014**, *1*, 1010. (g) Rouquet, G.; Blakemore, D. C.; Ley, S. V.; *Chem. Commun.* **2014**, *50*, 8908.
- Taylor, R. D.; MacCoss, M.; Lawson, A. D. G., *J. Med. Chem.* **2014**, *57*, 5845.
- (a) Ullrich, T.; Krich, S.; Mereiter, K.; Anderson, D. J.; Meyer, M. D.; Pyerin, M. *J. Med. Chem.* **2002**, *45*, 4047. (b) Lawson, E. C.; Hoekstra, W. J.; Addo, M. F.; Andrade-Gordon, P.; Damiano, B. P.; Kauffman, J. A.; Mitchell, J. A.; Maryanoff, B. E. *Bioorg. Med. Chem. Lett.* **2001**, *11*, 2619. (c) Kawata, S.; Ashzawa, S.; Hiram, M. *J. Am. Chem. Soc.* **1997**, *119*, 12012. (d) Rico, J. G.; Lindmark, J. R.; Bovy, P. R. *J. Org. Chem.* **1993**, *58*, 7948. (e) Aoki, M.; Ohtsuka, T.; Yamada, M.; Ohba, Y.; Yoshizaki, H.; Yasumo, H.; Sano, T.; Seto, H. *J. Antibiot.* **1991**, *44*, 582. (f) Atta-ur-Rahman; Shahwar, D.; Choudhary, M. I.; Sener, B.; Toker, G.; Baser, K. H. C. *Phytochemistry* **1999**, *50*, 333. (g) Husain, I.; Saquib, M.; Bajpai, V.; Kumar, B.; K. Shaw, A. K. *J. Org. Chem.* **2011**, *76*, 8930. (h) Bagley, M. C.; Dale, J. W.; Merritt, E. A.; Xiong, X. *Chem. Rev.* **2005**, *105*, 685. (i) Nicolaou, K. C.; Safina, B. S.; Zak, M.; Lee, S. H.; Nevalainen, M.; Bella, M.; Estrada, A. A.; Funke, C.; Zécéri, F. J.; Bulat, S. *J. Am. Chem. Soc.* **2005**, *127*, 11159.
- (a) Haas, J.; Piguel, S.; Wirth, T. *Org. Lett.* **2002**, *4*, 297. (b) Brunner, H.; Markus, N. *Monatsh. Chem.* **2002**, *133*, 115. (c) Chelucci, G.; Pinna, G. A.; Saba, A. *Tetrahedron: Asymmetry* **1997**, *8*, 2571. (d) Canary, J. W.; Allen, C. S.; Castagnetto, J. M.; Wang, Y. *J. Am. Chem. Soc.* **1995**, *117*, 8484. (e) Chelucci, G.; Conti, S.; Falorni, M.; Giacomelli, G. *Tetrahedron* **1991**, *38*, 8251. (f) Brunner, H.; Heinrich, F. *J. Organomet. Chem.* **1987**, *335*, 1. (g) Zai, S.; Gao, H.; Huang, Z.; Hu, H.; Wu, H.; Wu, Q.; *ACS Catal.* **2012**, *2*, 433. (h) Baratta, W.; Benedetti, F.; Del Zotto, A.; Fanfoni, L.; Felluga, F.; Magnolia, S.; Putignano, E.; Rigo, P.; *Organometallics* **2010**, *29*, 3563. (i) Cheng, Y. Q.; Bian, Z.; Kang, C. Q.; Guo, H. Q.; Gao, L. X. *Tetrahedron: Asymmetry* **2008**, *19*, 1572. (j) Baratta, W.; Bosco, M.; Chelucci, G.; Del Zotto, A.; Siega, K.; Toniutti, M.; Zangrando, E.; Rigo, P.; *Organometallics* **2006**, *25*, 4611. (k) Yano, T.; Tanaka, R.; Nishioka, T.; Kinoshita, I.; Isobe, K.; Wright, L. J.; Collins, T. J. *Chem. Commun.* **2002**, 1396. (l) Rowland, J. M.; Olmstead, M. M.; Mascharak, P. K. *Inorg. Chem.* **2002**, *41*, 1545.
- Nguyen, T.; Shapiro, D. A.; George, S. R.; Setola, V.; Lee, D. K.; Cheng, R.; Rauser, L.; Lee, S. P.; Lynch, K. R.; Roth, B. L.; O'Dowd, B. F. *Mol. Pharmacol.* **2001**, *59*, 427.
- Eccles, R.; van Cauwenberge, P.; Tetzloff, W.; Borum, P. *J. Pharm. Pharmacol.* **1995**, *47*, 990.
- (a) Harikrishnan, L.S.; Finlay, H.J.; Qiao, J. X.; Kamau, M. G.; Jiang, J.; Wang, T. C.; Li, J.; Cooper, C.B.; Poss, M.A.; Adam, L.P.; Taylor, D.S.; Chen, A. Y. A.; Yin, X.; Sleph, P.G.; Yang, R.Z.; Sitkoff, D.F.; Galella, M.A.; Nirschl, D.S.; Van Kirk, K.; Miller, A. V.; Huang, C.S.; Chang, M.; Chen, X. Q.; Salvati, M. E.; Wexler, R. R.; Lawrence, R. M.; *J. Med. Chem.* **2012**, *55*, 6162. (b) Miller, M.M.; Liu, Y.; Jiang, J.; Johnson, J.A.; Kamau, M.; Nirschl, D. S.; Wang, Y.; Harikrishnan, L.; Taylor, D. S.; Chen, A.Y.A.; Yin, X.; Seethala, R.; Peterson, T. L.; Zvyaga, T.; Zhang, J.; Huang, C. S.; Wexler, R. R.; Poss, M. A.; Lawrence, R. M.; Adam, L. P.; Salvati, M. E.; *Biorg. Med. Chem. Lett.* **2012**, *22*, 6503.

- <sup>8</sup> Robak, M. T.; Herbage, M. A.; Ellman, J. A. *Chem. Rev.* **2010**, *110*, 3600.
- <sup>9</sup> (a) Shaw, A. W.; deSolms, S. J. *Tetrahedron Lett.* **2001**, *42*, 7173. (b) Spero, D. M.; Kapadia, S. R. *J. Org. Chem.* **1997**, *62*, 5537. (c) Clayden, J.; Hennecke, U. *Org. Lett.* **2008**, *10*, 3567.
- <sup>10</sup> Clayden, J. *Organolithiums: Selectivity for Synthesis*; Pergamon: Oxford, 2002.
- <sup>11</sup> (a) Clayden, J.; Dufour, J.; Grainger, D. M.; Helliwell, M. *J. Am. Chem. Soc.* **2007**, *129*, 7488. (b) Vincent, M. A.; Maury, J.; Hillier, I. H.; Clayden, J. *Eur. J. Org. Chem.* **2015**, *5*, 953.
- <sup>12</sup> Clayden, J. In *Lithium Compounds in Organic Synthesis*; Luisi, R., Capriati, V., Eds.; Dearomatization and Aryl Migration in Lithium Chemistry; 2014.
- <sup>13</sup> Kavirayani, R. P.; Omkar R. *Tetrahedron* **2013**, *69*, 8422.
- <sup>14</sup> Xing-Wen, S.; Min L.; Ming-Hua X.; Guo-Qiang L. *Org. Lett.* **2008**, *10*, 1259.
- <sup>15</sup> Chelucci, G.; Baldino, S.; Chessa, S.; Pinna, G. A.; Soccolini, F. *Tetrahedron: Asymmetry* **2006**, *17*, 3163.
- <sup>16</sup> Fournier, A. M.; Brown, R. A.; Farnaby, W.; Miyatake-Ondoabazal, H.; Clayden, J. *Org. Lett.* **2010**, *12*, 2222.
- <sup>17</sup> Zimmerman, H. E.; Singer, L.; Thyagarajan, B. S. *J. Am. Chem. Soc.* **1959**, *81*, 108.
- <sup>18</sup> Clayden, J.; Donnard, M.; Lefranc, J.; Minassi, A.; Tetlow, D. J. *J. Am. Chem. Soc.* **2010**, *132*, 6624.
- <sup>19</sup> (a) Wiklund, T.; Olsson, S.; Lennartson, A. *Monatsh. Chem.* **2011**, *142*, 813. (b) Pratt, L. M.; Khan, I. M.; *THEOCHEM* **1996**, *367*, 33. (c) Holger, O.; Pieper, U.; Leusser, D.; Flierler, U.; Henn, J.; Stalke, D. *Angew. Chem. Int. Ed.* **2009**, *48*, 2978. (d) Pieper, U.; Stalke, D.; *Organometallics* **1993**, *12*, 1201.
- <sup>20</sup> (a) Maury, J.; Clayden, J. *J. Org. Chem.* **2015**, *80*, 10757. (b) Atkinson, R. C.; Fernández-Nieto, F.; Mas Roselló, J.; Clayden, J. *Angew. Chem. Int. Ed.* **2015**, *54*, 8961.
- <sup>21</sup> Tomohara, K.; Yoshimura, T.; Hyakutake, R.; Yang, P.; Kawabata, T. *J. Am. Chem. Soc.* **2013**, *135*, 13294.
- <sup>22</sup> Atkinson, R. C.; Leonard, D. J.; Maury, J.; Castagnolo, D.; Volz, N.; Clayden, J. *Chem. Commun.* **2013**, *49*, 9734.
- <sup>23</sup> The er of amines **16a-f**, like those of **12b,d** was not determined, but was assumed to be the same as the parent urea.
- <sup>24</sup> Tait, M.; Donnard, M.; Minassi, A.; Lefranc, J.; Bechi, B.; Carbone, G.; O'Brien, P.; Clayden, J. *Org. Lett.* **2013**, *15*, 34.
- <sup>25</sup> (a) MacLellan, P.; Clayden, J. *Chem. Commun.* **2011**, *47*, 3395. (b) Mingat, G.; MacLellan, P.; Laars, M.; Clayden, J. *Org. Lett.* **2014**, *16*, 1252. (c) Mingat, G.; McDouall, J. J. W.; Clayden, J. *Chem. Commun.* **2014**, *50*, 6754.
- <sup>26</sup> Clayden, J.; Farnaby, W.; Grainger, D. M.; Hennecke, U.; Mancinelli, M.; Tetlow, D. J.; Hillier, I. H.; Vincent, M. A. *J. Am. Chem. Soc.* **2009**, *131*, 3410.
- <sup>27</sup> (a) Bach, R.; Clayden, J.; Hennecke, U. *Synlett* **2009**, 421. (b) Tait, M. B.; Ottersbach, P. A.; Tetlow, D. J.; Clayden, J. *Org. Process Res. Dev.* **2014**, *18*, 1245. (c) Tait, M. B.; Butterworth, S.; Clayden, J. *Org. Lett.* **2015**, *17*, 1236.
- <sup>28</sup> (a) Tetlow, D. J.; Hennecke, U.; Raftery, J.; Waring, M. J.; Clarke, D. S.; Clayden, J. *Org. Lett.* **2010**, *12*, 5442. (b) Tetlow, D. J.; Vincent, M. A.; Hillier, I. H.; Clayden, J. *Chem. Commun.* **2013**, *49*, 1548.
- <sup>29</sup> Clayden, J.; Stimson, C. C.; Keenan, M.; Wheatley, A. E. H. *Chem. Commun.* **2004**, *2*, 228.
- <sup>30</sup> (a) Grainger, D. M.; Smith, A. C.; Vincent, M. A.; Hillier, I. H.; Wheatley, A. E. H.; Clayden, J. *Eur. J. Org. Chem.* **2012**, *4*, 731. (b) Lefranc, J.; Fournier, A. M.; Mingat, G.; Herbert, S.; Marcelli, T.; Clayden, J. *J. Am. Chem. Soc.* **2012**, *134*, 7286.